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RECONNAISSANCE OF THE QUALITY OF SURFACE WATER IN THE

UPPER VIRGIN RIVER BASIN, UTAH, ARIZONA, AND NEVADA, 1981-82

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Prepared by the United States Geological Survey in cooperation with the Utah Department of Natural Resources Division of Water Rights

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CONVERSION FACTORS

For readers who prefer to use metric units, conversion factors for inchpound units used in this report are listed below:

Multiply inch-pound units	By	<u>To obtain metric units</u>
acre	0.4047	square hectometer
	0.004047	square kilometer
acre-foot	0.001233	cubic hectometer
	1233	cubic meter
cubic foot per second	0.02832	cubic meter per second
cubic foot per second-day	0.02832	cubic meters per second-day
inches	2.54	œntimeter
	25.40	millimeter
foot	0.3048	meter
foot per mile	0.1894	meter per kilometer
gallon per minute	0.06308	liters per second
mile	1.609	kilameter
square mile	2.590	square kilometer
ton per day	0.9072	metric ton per day megagram per day
ton per day per square mile	0.3502	megagrams per day per square kilometer
	Temperature	

degree Fahrenheit $^{O}C=5/9$ $^{(O}F=32)$ degree Celsiusdegree Celcius $^{O}F=9/5$ $^{(O}C+32)$ degree Fahrenheit

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RECONNAISSANCE OF THE QUALITY OF SURFACE WATER IN THE

VIRGIN RIVER BASIN, UTAH, ARIZONA, AND NEVADA, 1981-82

by George W. Sandberg and LaVerne G. Sultz Hydrologists, U.S. Geological Survey

ABSTRACT

The Virgin River drainage from headwaters in Utah to Littlefield, Arizona, has an area of about 5,090 square miles in southwestern Utah, northwestern Arizona, and southeastern Nevada. In this area the river is about 110 miles long and receives water from nine major tributaries. Two tributaries, Fort Pierce Wash and Beaver Dam Wash, are ephemeral with flow occurring only from irrigation-return, floods or snowmelt, and springs in short reaches.

Outcropping geologic formations that affect water quality in the basin are fine-grained clastic and carbonate rocks of Mesozoic age.

Tributary inflow to the Virgin River during the study generally contained smaller dissolved solids concentrations than did the river and significantly diluted dissolved-solids concentrations in the river during low flow. Tributaries generally contained larger dissolved-solids concentrations than did the river during high flow, but the proportion of water from triburaries was smaller during high flow and dilution effect was relatively small.

La Verkin Hot Springs enters the river at Hurricane fault and contributes a flow of about 11 cubic feet per second. Dissolved-solids concentration of this springs is nearly 10,000 milligrams per liter with sodium and chloride constituting the major ions. Approximately 109,000 tons of dissolved solids flow from the springs annually.

During low flow, dissolved-solids concentrations in the Virgin River ranged from only 56 milligrams per liter in the North Fork Virgin River at Cascade Spring to 603 milligrams per liter upstream from La Verkin Hot Springs, and from 2,760 milligrams per liter downstream from the hot springs to 2,620 milligrams per liter at Littlefield, Arizona. During high flow, the dissolved-solids concentration upstream from La Verkin Hot Springs was 277 milligrams per liter and the range downstream from the springs was 492 to 1,120 milligrams per liter. Dissolved-solids concentrations were maximum during low flow and minimum during high flow at all sites except on some small streams near triburary headwaters. Boron concentration was less than the tolerance level of all crops upstream from La Verkin Hot Springs and was more than the tolerance level of many crops downstream from the springs.

Sodium hazard was low to medium except just downstream from La Verkin Hot Springs; the salinity hazard generally was low to high upstream from La Verkin Hot Springs and high to very high downstream from the springs.

Sediment loads ranged from 0.13 to 2,555 tons per day for 25 samples collected in August 1981 and from 0.55 to 3,582 tons per day for 14 samples collected in May 1982. These loads reflect stable stream conditions during sampling periods. Loads during flood flows have been as much as 1,930,000

tons per day for the period of record. Largest sediment loads were in the southwestern part of the basin, and largest sediment loads per square mile of drainage area were in the northeastern part of the basin where gradients are steeper.

Dissolved-oxygen concentrations ranged from 5.2 to 12.5 milligrams per liter except immediately downstream from thermal springs. Manganese concentrations were in excess of recommended limits for drinking water at several sites. Data were insufficient to identify sanitary problems but available data indicate that conditions may be degraded downstream from livestock areas. Pesticide concentrations were minimal at the few sites sampled.

INTRODUCT ION

This report on the quality of surface water in the Virgin River basin from the headwaters near Navajo Lake, Utah, to Littlefield, Ariz., was prepared by the U.S. Geological Survey in cooperation with the Division of Water Rights, Utah Department of Natural Resources. The objectives of the study leading to the report were to obtain information on general chemical characteristics of surface water and to determine effects of the natural environment and water use on these characteristics. The scope of the study did not include an intensive investigation of the effects of man's activities on water quality.

Methods of Investigation

Water-quality data were obtained 1 to 5 times at 74 sites in the Virgin River basin between August, 1981 and September 1982. Fifteen additional sites were dry during all the sampling periods. The sites were numbered in consecutive upstream order, beginning at Littlefield, Arizona. The locations are shown on plate 1. Thirty-one sites were designed as key sites, and they were sampled during each of the five sampling periods, if accessible. The key sites included active gaging stations, locations at, upstream from, or downstream from major tributaries, and other locations of probable waterquality significance.

The concentrations of selected trace elements were determined semiguantitatively once at 23 sites and quantitatively once at 6 sites. The concentrations of pesticides in stream-bottom sediments were determined once at 4 sites.

Suspended-sediment samples were collected at 25 sites in August 1981, when flows were as low as 2 ft³/s and at 14 sites in May 1982, when flows were as high as 600 ft³/s. Sampling periods were selected when streamflow was most uniform, in order to define chemical characteristics of the river system. Sediment samples, therefore, represent only loads during uniform flow conditions.

Most streamflow measurements were made with fewer sections for velocity and depth determination than are used in standard stream-gaging procedures. This procedure was used in order to minimize the time needed for the large numbers of samples and measurements. Comparison of results of the two procedures on selected measurements were within 10 percent. The U.S. Geological Survey began streamflow measurements at site 32 in 1909 in cooperation with the Utah State Engineer. The Geological Survey currently (1983) operates 16 stream-gaging stations in the study area and an additional 15 have been operated in the past.

Previous Studies

Patterson and Somers (1966) reported on the magnitude and frequency of floods in the Virgin River, and the U.S. Army Corps of Engineers (U.S. Department of Defense, 1973) published flood-plain information for the Virgin River and Fort Pierce Wash. Geological Survey studies concerned with water quality resulted in reports on the thermal springs of Utah (Mundorff, 1970), the disposition of water seeping from Navajo Lake (Wilson and Thomas, 1964), and a map showing the general chemical quality of surface water in parts of the study area (Price, 1980). Cordova, Sandberg, and McConkie (1972) and Cordova (1978 and 1981) reported on the chemical quality of ground water in the central Virgin River basin. Water-quality data, including trace metals and bacteria, were collected by the Geological Survey at sites 1, 17, and 32 prior to this study, the earliest being in 1949 at site 1. Water-quality and sediment data have been published annually in several series of reports of the Geological Survey (1974 and 1982).

Additional reports concerned with water quality include those by Deacon and Holden (1977), Vaughn Hansen Associates (1977), and Gebhardt (1977), and the Five County Association of Governments (1977). The U.S. Soil Conservation Service (1981) and the U.S. Bureau of Reclamation (1982) completed reports on the lower Virgin River, which includes part of the study area. Trudeau (1979) and Moore (1969) reported on Littlefield Springs and surface-water flow in the vicinity of Littlefield. The Utah Water Research Laboratory (1974) reported on planning for water quality in the Virgin River system, and Goode (1964) reported on the East Fork Virgin River.

HYDROLOGIC SETTING

Surface Drainage

The Virgin River basin (pl. 1) has an area of about 5,100 square miles, of which 3,000 are in Utah, 1,700 in northwestern Arizona, and 400 in southeastern Nevada. The river length is approximately 110 miles from its origin near Navajo Lake, Utah, to littlefield, Arizona. Nine major tributaries enter the river in the study area. They are East Fork Virgin River, North Fork Virgin River, North Creek, La Verkin Creek, Ash Creek, Leeds Creek, Fort Pierce Wash, Santa Clara River, and Beaver Dam Wash. Numerous small perennial and ephemeral tributaries enter the Virgin River and its major tributaries.

Flow in the North Fork Virgin River begins at Cascade Springs, which is partly sustained by water from Navajo Lake (Wilson and Thomas, 1964). Fort Pierce Wash, entry point for virtually all flow from Arizona, and Beaver Dam Wash, entry point for all flow from Nevada, usually are dry or have small flows. These washes, however, can have large sediment-laden flows during flash floods. The Utah section of the study area contains about 15 reservoirs, most of which provide a small volume of water for irrigation. The three largest reservoirs are Kolob (6,900 acre-feet) on Kolob Creek about 10 miles southeast of Kanarraville; Gunlock (15,000 acre-feet) on the Santa Clara River about 1 mile south of Gunlock; and Ash Creek (12,000 acre-ft) on North Ash Creek about 9 miles south of Kanarraville. Capacity records are kept only for Ash Creek Reservoir where water is not usually released because most of it seeps into the porous basalt that underlies the reservoir. No reservoirs that are larger than stock ponds exist in the Arizona and Nevada sections of the study area.

Geology

The surface rocks in the upper Virgin River basin range in age from Precambrian to Quaternary. Similar geologic units have been grouped together for simplicity in plate 1. Groups that probably affect water quality most are the fine-grained clastic rocks of Mesozoic age, including the Tropic Shale and the Kayenta, Chinle, and Moenkopi Formations; and the carbonate rocks of Paleozoic age, including the Kaibab Limestone; Toroweap Formation; Callville, Redwall, Temple Butte, and Muddy Peak Limestones; Nopah and Bonanza King Formations; and the Muav Limestone. Although the rocks in the basin are extensively faulted, only the largest fault--Hurricane fault-- which is defined mainly by the Hurricane Cliffs, is shown on plate 1.

Topography

The drainage basin generally slopes toward the southwest in Utah, the northwest in Arizona, and the south in Nevada. Altitudes range from above 10,000 feet near the headwaters to below 1,900 feet at Littlefield. Some mountain areas have sheer slopes of several hundred feet (fig. 1). Steep stream gradients result in the erosion of large quantities of sediment, significantly affecting water quality. The terrain changes progressively downstream from high mountains and terraces to mesas and low desert land. Much of the area is virtually inaccessible. Major topographic features within the basin are Beaver Dam, Bull Valley, and Pine Valley mountains, Hurricane Cliffs, Vermillion Cliffs, and Kolob Terrace.

<u>Soils</u>

Most soils within the basin are sandy, sandy loam, or sandy loam with gravel, and range from shallow to deep (Mortensen and others, 1977; Richmond and Richardson, 1974) soils map. Most soils are well drained. Infertile soil in much of the basin supports little vegetation, particularly in desert areas where precipitation is meager. Some areas have virtually no soil and no vegetation.

Land Use

Recreation is a major land use. Zion National Park, several State parks, recreational areas, and mountain subdivisions for summer homes are located within the study area, mostly in Utah. Zion National Park is visited by about 1.5 million people annually including as many as 300,000 a month during the summer.



Figure 1.—Deep Creek drainage area, part of North Fork Virgin River and Zion Canyon area. Several sampling sites are located within the plateau area and in the distant canyon. Deficiency of water precludes agricultural development in many parts of the basin. Main agricultural areas are in the vicinity of St. George and Hurricane. Limited farming also is done in the smaller valleys.

Much of the basin is used for rangeland, but sparse range growth generally limits grazing. Most of the grazing is in the forested areas, where small areas have been reserved.

<u>Climate</u>

Normal annual precipitation in the basin ranges from more than 40 inches in the mountains to less than 8 inches in the low desert areas (pl. 2). Rain and melting snow in the mountains during spring and early summer provide water for downstream use. The small proportion of precipitation that falls on desert areas (mostly rain) either sinks into the porous soil or runs off as flash floods.

Temperatures generally range from 24 to -29 ^OCelsius near the headwaters; the mountain areas have fewer than 90 frost-free days. Temperatures generally range from 43 to -7 ^OCelsius in the desert areas; these areas have about 200 frost-free days (National Oceanic and Atmospheric Administration, 1981).

Water Sources and Use

Base flow of the Virgin River originates mainly from springs and seeps. Higher flows during spring and early summer result from snowmelt. Flash floods contribute little to total annual flow. Virtually all flow upstream from Littlefield originates in Utah. The small quantity of flow that occurs in Arizona and Nevada usually appears and disappears intermittently in streambeds, is ponded for stockwater, or is used for limited irrigation near the source.

Most use of water from the perennial streams is for irrigation. Water is diverted from the main channel of the Virgin River and tributaries along the entire course of the river. Largest diversions are to the Hurricane and La Verkin Canals, about 4 miles northeast of Hurricane, and the Washington Canal, about 7 miles east of St. George. Ground water is pumped for irrigation, particularly in the St. George-Hurricane area, and irrigation runoff adds to river flow downstream from these irrigated areas.

Phreatophytes along the streams consume water. The phreatophyte growth is mainly in downstream reaches where gradients are relatively flat and more soil exists.

CLASSIFICATION OF WATER FOR PUBLIC SUPPLY AND IRRIGATION

"The National Interim Primary Drinking Water Regulations*** were promulgated on December 24, 1975, in accordance with the provisions of the Safe Drinking Water Act (Public Law 93-523) *** These regulations become effective on June 24, 1977, and became in essence the standards by which all public drinking water supplies are judged" (U.S. Environmental Protection Agency 1976a, preface). The following table lists maximum contaminant levels for inorganic chemicals other than fluoride. The term "maximum contaminant level" is defined as the "maximum permissible level of a contaminant in water which is delivered to the free flowing outlet of the ultimate user of a public water system" (U.S. Environmental Protection Agency, 1976a, p. 5).

Contaminant	Level (milligrams per liter)
Arsenic Barium Cadmium Chromium Lead Mercury Nitrate (as N) Selenium Silver	1.0 .010 .05 .05 .002 10.0 .01

When annual average of the maximum daily air temperature for the location in which the community water system is situated is the following, maximum contaminant levels (approved limits) and other recommended control limits for fluoride are (U.S. Environmental Protection Agency, 1976a, p. 5):

		Mil	ligrams pe	r liter	
Tempe	rature	Recommended of for fluoride of			
Degrees	Degrees	Lower	Optimum	Upper	Approved
Fahrenheit	Celsius	Lower	Optimum	Upper	limit
53.7 and below	12.0 and below	1.1	1.2	1.3	2.4
53.8 to 58.3	12.1 to 14.6	1.0	1.1	1.2	2.2
58.4 to 63.8	14.7 to 17.6	.9	1.0	1.1	2.0
63.9 to 70.6	17.7 to 21.4	.8	.9	1.0	1.8
70.7 to 79.2	21.5 to 26.2	.7	.8	.9	1.6
79.3 to 90.5	26.3 to 32.5	.6	.7	.8	1.4

Hardness of water is conventionally expressed in all water analyses made in the United States in terms of an equivalent quantity of calcium carbonate (CaO_3). Some such convention is needed for hardness because this is a property imparted by several different cations, which may be present in varying proportions; however, actual presence of the indicated number of milligrams per liter in the form of $CaCO_3$ certainly should not be assumed (Hem, 1970, p. 84).

In practical water analysis, hardness is computed by multiplying the sum of milliequivalents per liter of calcium and magnesium by 50. Hardness value resulting generally is entitled "hardness as CaO_3 "....or "total hardness". If hardness exceeds alkalinity (in milligrams per liter of CaCO₃ or other equivalent units), the excess is termed "noncarbonate hardness".... (Hem, 1970, p. 224-225).

Durfor and Becker (1964, p. 27) use the following classification for hardness:

Hardness range (milligrams per liter of calcium carbonate)	Description	
0-60	Soft	
61-120	Moderately hard	
121-180	Hard	
More than 180	Very hard	

The U.S. Environmental Protection Agency produced a group of "Quality Criteria for Water" to provide a basis for judgement, other than regulatory use, for several programs that are associated with water quality. Following is a list of selected recommended limits for drinking-water supplies (U.S. Environmental Protection Agency, 1976b):

Constituent	Concentration		
	Milligrams per liter	Micrograms per liter	
Beryllium		100	
Chloride	250		
Copper		1,000	
Dissolved oxygen	5		
Iron	2000 March 1990	300	
Manganese		50	
Sulfate	250		
Dissolved solids	500		

A classification for the dissolved-solids hazard in irrigation waters has been prepared by the U.S. Environmental Protection Agency (1976b, table 16). This classification is shown below:

Dissolved-solids concentration milligrams per liter	Effects or limitation
Less than 500	Usually none
500-1,000	Can be deterimental to some sensitive crops
1,000-2,000	May have adverse effects on many crops and require careful management
2,000-5,000	Can be used for tolerant plants on premeable soils, requires careful management practices

Salinity and sodium hazards of water used for irrigation are classified using a diagram developed by the U.S. Salinity Laboratory Staff (1954, p. 80). This diagram shows electrical conductivity of the water (salinity hazard) which is an indication of dissolved-solids concentration; and sodiumabsorption ratio or SAR (sodium hazard), which is the relationship of sodium (Na⁺), calcium (CA⁺⁺), and magnesium (Mg⁺⁺) ions expressed by the equation:

$$\frac{\text{Na}^+}{\text{Ca}^+ \text{Mg}^{++}}$$

The assumption is made that water will be used under average soil and drainage conditions. If a large deviation from average conditions occurred, water could become unsuitable for use even though under average conditions it would be suitable for irrigation.

Boron may be a limiting factor in irrigation waters and is, therefore, considered in assessing water quality. The U.S. Environmental Protection Agency (1976b, p. 25) recommends a maximum concentration of 750 micrograms per liter of boron in water for the most boron-sensitive plants. Hem (1970, p. 329) rates irrigation water for various crops on the basis of boron concentrations in the water as shown in the following table:

		Borc	on (milligrams per	liter)
Class of water			Crops	
Rating	Grade	Sensitive	Semitolerant	Tolerant
1	Excellent	0.33	0.67	1.00
2	Good	.33 to .67	.67 to 1.33	1.00 to 2.00
3	Permissible	.67 to 1.00	1.33 to 2.00	2.00 to 3.00
4	Doubtful	1.00 to 1.25	2.00 to 2.50	3.00 to 3.75
5	Unsuitable	1.25	2.50	3.75

Boron-sensitive crops include pears, apples, cherries, pecans, peaches, and apricots. Semitolerant crops include potatoes, barley, wheat, corn, milo, and oats. Tolerant crops include alfalfa and sugar beets.

CHEMICAL QUALITY OF SURFACE WATER

Water types are characterized in this report using an arbitrary nomenclature (Davis and DeWiest, 1966, p. 119). Major ions present as less than 20 percent of the total milliequivalents per liter of cations or anions are not used to name the water type. If any ion represents more than 60 percent of the total milliequivalents per liter of either cations or anions this ion is used alone to represent the dominant ion type. In mixed water types, ions present in greater than 20 percent but less than 60 percent of the cations or anions are listed in the order of their abundance. For example, water at site 2 during May 1982 had a 47 percent calcium, 28 percent sodium, 23 percent magnesium, and 2 percent potassium of cations and 52 percent bicarbonate, 37 percent sulfate, and 10 percent chloride of anions. This would be a calcium sodium magnesium bicarbonate sulfate water type. Bicarbonate is represented by alkalinity in tables 7 and 8 (back of report).

Streams in the Virgin River drainage flow from areas that are considerably different from each other in geology, land use, vegetation, altitude, and climate. Water quality is measurably affected by these differences. Solutes are determined by rock and soil composition, climate, biological effects of plants and animals, and water management and use as the water flows downstream. Water in the upstream reaches has relatively small concentrations of dissolved solids because much of the flow is derived from rainfall and snowmelt and has been in contact with soil and rocks for relatively short periods. Water-quality changes for two sampling periods are shown in plate 3. Changes in dissolved-solids concentrations were generally gradual between sites and changes in patterns in the illustrations show the approximate location of the change in limits as represented by the patterns. Data were not available where no pattern is shown on the stream.

Classification of water for irrigation at selected sites indicates downstream change in quality of the water for irrigation (figs. 2 and 3).

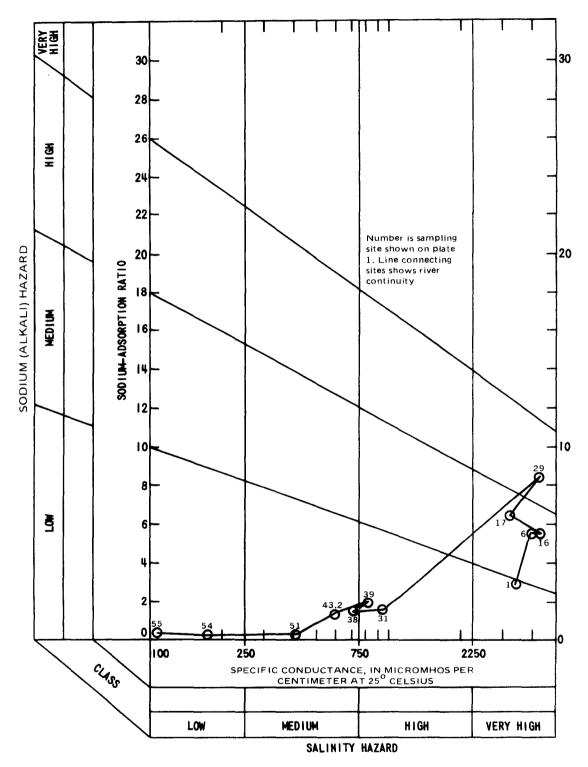


Figure 2.-Diagram showing classification of water for irrigation at selected sites on the Virgin River during low flow, August 17-31, 1981.

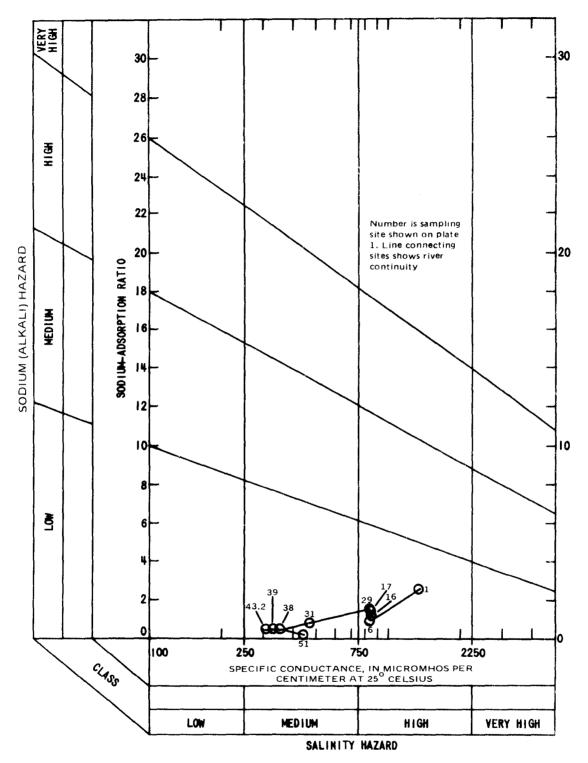


Figure 3.-Diagram showing classification of water for irrigation at selected sites on the Virgin River during high flow, May 3-11, 1982.

Changes in concentrations along the river during the five sampling periods are shown in figure 4. Dissolved-solids concentrations were affected significantly by the quantity of flow during various periods. Flows and concentrations of various constituents for sampling periods are shown in table 8. Downstream variations in the water quality are discussed in the following sections.

Dissolved Solids and Major Ions

North Fork Virgin River Drainage Area

Altitudes along the North Fork Virgin River range from about 8,900 feet at Cascade Spring to 3,800 feet at the confluence with the East Fork Virgin River. Average gradient of the river through this section is about 190 feet per mile although the gradient in some upstream reaches is much steeper.

The North Fork Virgin River begins at Cascade Spring, which emerges from the Tertiary Wasatch Formation about 6,400 feet south of Navajo Lake. Water sinks into porous basalt at the east end of Navajo Lake (fig. 5) and contributes much of the flow from Cascade Spring (Wilson and Thomas, 1964, p. 12, 13). Flow from Cascade Spring (site 54) during the investigation was lowest in October 1981 when the sink area of Navajo Lake had been dry for about 2 months and highest in August 1982 when water had covered the sink area since early summer. Dissolved-solids concentrations were largest; 139 milligrams per liter during the low October flow, but were smallest, 56 milligrams per liter, during the intermediate flow of August 1981 (table 8). In both cases the water was a calcium magnesium bicarbonate type.

Between Cascade Spring (site 54) and the head of Zion Narrows (site 51) many seeps, ephemeral streams, and small spring-sustained perennial streams enter the North Fork Virgin River. Deep snow packs and summer thunderstorms contribute most of the flow with discharge from springs and seeps comprising the base flow. Some water is diverted for irrigation of meadows, but most returns to the stream on the surface or as seepage. The area is used extensively for grazing and recreation during the summer. The dissolvedsolids concentrations were between 222 and 278 milligrams per liter at site 51. Water type remained the same as at Cascade Spring.

Zion Narrows and Deep Creek area, characterized by steep canyons and sheer cliffs, was inaccessible for collecting water samples between sites 44.1, 45, 50 or 51 and site 43.2 (pl. 1). Within this area, Deep Creek, Kolob Creek, and Orderville Gulch enter the North Fork Virgin River. Terrain and water use of Deep and Kolob Creeks are similar to those of the North Fork Virgin River upstream from site 51 and water samples collected near their headwaters indicate their quality is similar to that of the North Fork. Orderville Gulch originates at a lower altitude and in a different rock formation with a resulting difference in water quality.

The Dakota Sandstone and Tropic Shale of Cretaceous age, coal-bearing formations, cropout between sites 51 and 43.2. Coal in these formations could affect quality.

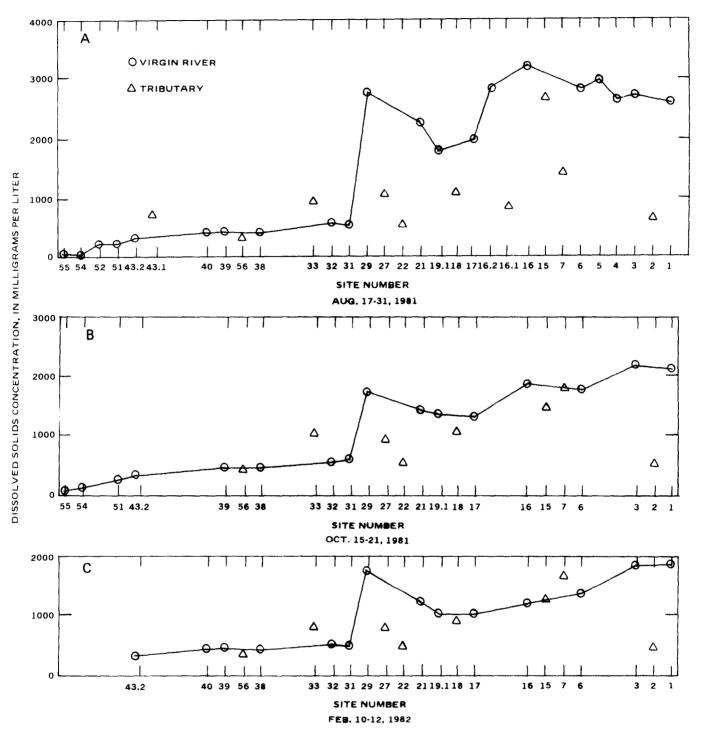


Figure 4.-Concentration of dissolved solids at selected sites.

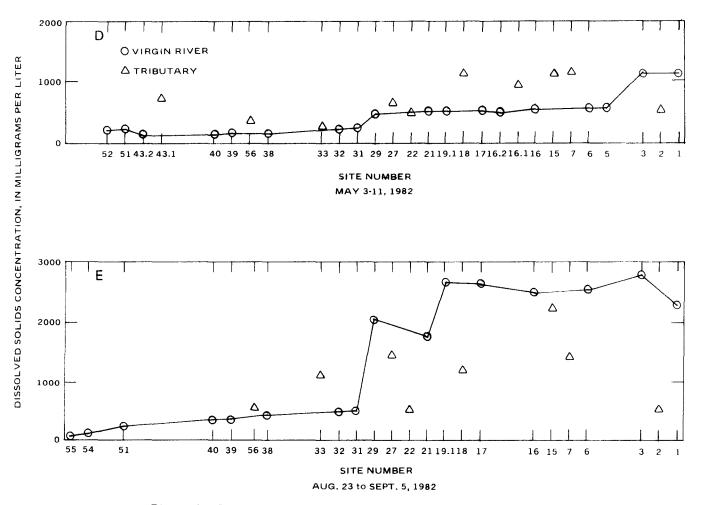


Figure 4.-Concentration of dissolved solids at selected sites-Continued.

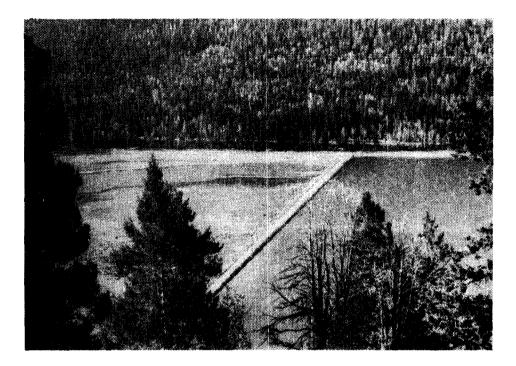


Figure 5.—Dike across east part of Navajo lake where water sinks into the lake bed, October 1982. Site 55 is at far end of dike.

Base flow through the Zion Narrows reach is composed of discharge from springs and seeps. Flow through this reach increased from 7.3 cubic feet per second at site 51 to 54 cubic feet per second at site 43.2 in October 1981, a typical period of low flow. Dissolved-solids concentration increased during all periods of low flow. Largest increases of sodium, sulfate, and chlorides changed the water to a calcium magnesium sodium bicarbonate sulfate type (table 7). Dissolved solids decreased during high flow in May 1982. Change in ions and dissolved solids are shown in plate 3.

Agricultural land along the North Fork Virgin River is located between Springdale, about 10 miles downstream from site 43.2, and site 39. This land is irrigated from small canals and ditches using water from the river, and runoff from irrigated areas returns to the river. The irrigation-return flow adds significantly to the mineral concentration of the river. Dissolvedsolids concentrations increased about 30 percent between sites 43.2 and 39 during low flow and about 15 percent during high flow. Largest ion increases were sodium, sulfate, and chloride. The water was a calcium sodium magnesium chloride bicarbonate sulfate type during low flow and a calcium bicarbonate type during high flow.

Downstream increase in mineral concentration (as shown by increasing specific conductance) is shown in figures 2 and 3. In August 1981, when discharge was low, there was an increase in specific conductance of the water, but little change in SAR between sites 55 and 51 (fig. 2). In May 1982, when discharge was high, specific conductance decreased between sites 51 and 43.2, and increased slightly between sites 43.2 and 39 (fig. 3). SAR remained virtually the same along the entire reach, showing the dilution effect of the larger flow.

Change in dissolved solids during the five sampling periods are shown in figure 4. Low-flows (fig. 4, a, b, c, e) had increases in dissolved-solids concentrations, and high-flows (fig. 4, d) had little change in the concentrations from site 52 to site 39.

East Fork Virgin River Drainage Area

Altitudes along the East Fork Virgin River range from about 7,200 feet at the headwaters to 3,800 feet at confluence with the North Fork. Average gradient through this reach is about 90 feet per mile. Terrain of the East Fork drainage is less rugged than that of the North Fork drainage, although steep canyons and high cliffs make the reach from Mt. Carmel Junction (site 60) to the mouth of East Fork (site 56) generally inaccessible. Much of this drainage area is used extensively for grazing.

The Tropic Shale and Dakota Sandstone, both with veins of coal cropout between Glendale and Zion National Park. Small active and abandoned mines have been developed throughout these formations.

Headwaters of the East Fork Virgin River originate from seeps and springs, mostly in grazed meadow areas. Geologic Formations (pl. 1) and soil types are similar to those in the headwaters of the North Fork Virgin River; however, the headwaters of the East Fork have lower altitudes and less precipitation than the headwaters of the North Fork. Dissolved-solids concentrations in the East Fork upstream from site 64 were between 200 and 350 milligrams per liter which are similar to those at sites 51 and 52 on the North Fork. The water was a magnesium calcium bicarbonate type.

Land in the vicinities of Glendale, Orderville, and Mt. Carmel is irrigated with water from the East Fork Virgin River, and most of the flow through these areas is used for irrigation. Water also is diverted from upstream reaches of the East Fork and tributaries to irrigate meadows and mountain pastures. Runoff from all irrigated areas returns to the streams.

Dissolved-solids concentrations increased about 3.5 times through the irrigated areas (site 64 to site 60) during the irrigation season. Dissolved-solids concentration decreased more than 100 percent between sites 60 and 56 in August 1981 because natural inflow within this reach diluted the irrigation runoff. Flow at site 56 was more than 10 times that at site 60 (table 8). Dissolved-solids concentration were 1 to 2 times less at site 64 than at site 60 during nonirrigation periods. During these periods the dissolved-solid concentration at site 60 was less, the flow was larger, and the percent increase of natural inflow between sites 60 and 56 was less, resulting in dissolved-solid concentration decreases of less than 50 percent. Greatest decreases were in calcium, magnesium, and sulfate. Changes in ion concentration and total dissolved solids along the East Fork for high and low flow are shown in plate 3.

Virgin River Drainage Area from the Confulence of the North and East Forks to La Verkin Hot Springs

Altitudes along this reach of the river range from about 3,800 feet at the confluence of the East and North Forks to about 3,200 feet at La Verkin Hot Springs. The gradient averages 38 feet per mile. Small ephemeral and perennial streams enter the river in this reach.

Dissolved-solids concentrations were smaller in the East Fork upstream from the confluence with the North Fork (site 56) than in the North Fork upstream from the confluence for all low-flow samples except August 23, 1982 (site 49). During high flow the concentrations upstream from the confluence were larger in the East Fork than in the North Fork (fig. 4). A slight decrease in specific conductance and SAR in August 1981 at site 38 caused by mixing the East Fork and North Fork waters is shown in figure 5. Virtually no change occurred in May 1982 (fig. 3) when flow was higher in the North Fork.

The river flows in a wide meandering channel composed mostly of unstable sand from the confluence (site 38) to site 32. High discharge including flash floods can readily change channel shape and alignment. Phreatophyte growth along the channel includes cottonwood trees, willows, and salt cedar (tamarisk). Water is diverted for irrigation in this area. Predominant constituents at site 32 for five analyses during the study period (table 8 back of report) and the average of seven analyses prior to the study period (table 9) showed calcium, sulfate, and bicarbonate to be the dominant ions.

North Creek (site 33) was the only perennial tributary in this reach large enough to sample. This inflow had nearly twice the dissolved-solids concentration of the river, but discharge was so small in comparison to river flow that the dissolved solids had little effect on the river quality. From site 32 to site 29 the river flows through a progressively deeper gorge where it erodes through higher terrain on the upthrown side of the Hurricane fault. River gradient is about the same along the entire reach between site 38 and 29 although the higher terrain gives the illusion of a steeper gradient between sites 32 and 29. The gorge is mostly in Kaibab Limestone of Permian age which crops out and forms cliffs in the vicinity of La Verkin Hot Springs. Water is diverted from the river into Hurricane and La Verkin Canals 2 miles upstream from La Verkin Hot Springs and within the gorge. Most river flow is diverted during periods of maximum irrigation. The largest quantity diverted during sampling periods of this study was about 60 percent of the total flow in August 1981.

Dissolved solids increased about 25 percent between sites 38 and 31 during low flow and about 35 percent during high flow. Specific conductance of the water increased proportinately and SAR increased slightly (figs. 2 and 3). Water along this reach generally was a calcium magnesium sodium bicarbonate sulfate type. Ion distribution at various sites is shown in plate 3.

Highly mineralized hot water from La Verkin Hot Springs (site 30) on the Hurricane fault enters the river through the bed and banks of the channel. Water temperatures in the different springs range from 38 to 42° Celsius. Inflow from the springs is about 11 cubic feet per second (Mundorff, 1970, p. 44) and constitutes a large percentage of the river discharge during low flow. Major ions in the spring water are sodium and chloride, with sodium more than double calcium and magnesium and chloride about double sulfate. Mundorff (1970, p. 46) reported that annual discharge of dissolved solids from the springs to be about the same as that for the entire Virgin River basin upstream from the springs. The U.S. Bureau of Reclamation (1979, p. 1) reported that the springs discharge 109,000 tons of dissolved solids annually.

Discharge from La Verkin Hot Springs (site 30) changed water quality in the Virgin River most significantly between sites 31 and 29 during the August 1981 sample period when flow was lowest, and least significantly in May 1982 when flow was highest of the sample periods. Change in ions and dissolvedsolids concentrations during these periods are shown in figures 3 and 4 and change in dissolved-solids concentrations for the five sampling periods are shown in figure 4. There was a large increase in specific conductance and SAR between sites 31 and 29 in August 1981 (fig. 2) when the spring flow was about 25 percent of total river flow. Increases were much smaller in May 1982 (fig. 3) when the spring flow was less than 2 percent of the river flow.

Boron concentrations increased from less than 100 micrograms per liter upstream from the springs to concentrations ranging from 150 to 1,300 micrograms per liter downstream from the springs. This general range persisted in the river as far downstream as Littlefield (table 3), and varied inversely with the quantity of water flowing in the river.

Mixing characteristics of La Verkin Hot Springs and Virgin River waters are shown in table 1. Data for the table were collected during low flow on September 22, 1982.

Virgin River Drainage Area from La Verkin Hot Springs to Santa Clara River

Altitudes range from 3,200 feet at La Verkin Hot Springs to 2,500 feet at Fort Pierce Wash with a gradient averaging 35 feet per mile.

Two main tributaries, La Verkin Creek and Ash Creek, enter the river from the north about 2 miles downstream from La Verkin Hot Springs. La Verkin Creek begins in the vicinity of Kolob Reservoir and drains mountainland along the east side of Hurricane fault. Outcroping geologic formations include sandstone, siltstone, and shale of Mesozoic age. Terrain is extremely rugged and the stream was inaccessible except in the downstream reach. One sampling site (27) was established at the mouth of the stream. Water at this site was a calcium magnesium sulfate bicarbonate type. Dissolved-solids concentrations ranged from 654 to 1,470 milligrams per liter. Sulfate was the ion of greatest concentration.

Ash Creek flows along the west side of Hurricane fault and is the main drainage for the east side of the Pine Valley Mountains. Geologic formations are mostly alluvium, basalt, and coarse-grained rocks. Ash Creek drainage is larger and streamflow generally is much larger than that of La Verkin Creek. Small tributaries, many ephemeral, from the Pine Valley Mountains, enter Ash Creek. Flow from snowmelt provided the only sample (site 25.1) from these tributaries. Kanarra Creek (site 26.1), a tributary in the northeastern part of the drainage, is completely diverted and used for irrigation in the Kanarraville area. This stream had a dissolved-solids concentrations of less than 300 milligrams per liter during low and high flows.

North and South Ash Creeks are the only perennial tributaries to Ash Creek. Flow from South Ash Creek is diverted for irrigation in the vicinity of Pintura. Water from North Ash Creek is impounded in Ash Creek Reservoir and mostly seeps into the basalt on which the reservoir is built. The reservoir was originally intended for irrigation but was not completed to specifications and is used mainly for flood control. Flow from Ash Creek Reservoir to Toquerville Springs, upstream from site 23, is from flash floods, snowmelt, or releases from Ash Creek Reservoir, all of which occur infrequently. All flow in Ash Creek at site 23 came from Toquerville Springs during the sample periods.

Toquerville Springs emerge from the streambed and banks of Ash Creek at the northern edge of Toquerville. Total flow from the springs, averages 20 cubic feet per second with flows from individual springs or seeps ranging to 4 cubic feet per second. Most of the water is diverted for irrigation between the springs and site 23. Origin of the springs is not known, but they are probably are sustained by underflow from Ash Creek or La Verkin Creek or both. Dissolved-solids concentrations (450 milligrams per liter) at site 23 were largest in the August 1981 sample. Proportionate ion concentrations were similar to those of La Verkin Creek with sulfate predominating. La Verkin Creek at site 27 and Ash Creek at site 22 had significantly smaller dissolved-solids concentrations than the Virgin River at site 29 during the periods of low flow and diluted the river flow to site 21 (pl. 3). Further downstream, decreases in the river salinity occurred at site 19.1 in August and October 1981, and February 1982, probably because of fresher ground-water inflow.

Gould Wash enters the Virgin River from the south about 3 miles downstream from Ash Creek and is dry except during flash floods and return flow from irrigation in the vicinity of Hurricane. Two samples collected at the mouth of the wash (site 20.1) had dissolved-solids concentrations of 567 and 369 milligrams per liter. Concentrations of both samples were much smaller than concentrations in the river.

Leeds Creek originates in the southern part of Pine Valley Mountains and flows generally south to the Virgin River. Dissolved-solids concentration was 175 milligrams per liter in the one sample obtained at the upstream site (site 19) on Leeds Creek and ranged from 901 to 1,200 milligrams per liter for the five samples at the downstream site (site 18). The large dissolved-solids increase in the stream probably was caused by return flow from irrigation and seepage through Moenkopi and Shinarump Formations that contain considerable soluble minerals. Increases in dissolved-solids concentrations in the Virgin River between sites 19.1 and 16.2 in August 1981, and between sites 21 and 17 in August 1982, probably were caused by irrigation return and inflow from Gould Wash and Leeds Creek. Predominant constituents at site 17 for five analyses during the study (table 4) and the average of five analyses prior to the study (table 9) were sodium, sulfate, and chloride.

River flow is diverted to the Washington Canal (site 16.2) during the irrigation season. Water is used to irrigate about 4,900 acres of the more than 5,300 acres of cropland in the St. George-Washington area (Utah Division of Water Resources, 1982, p. 104).

An estimated 25,000 to 35,000 acre-feet of ground water per year seeps into streams or is pumped from wells and flows into streams in the central Virgin River area mainly in the Hurricane to St. George area (Cordova and others, 1972, p. 17). Dissolved-solids concentrations of some of the pumped water are larger than those of river water, but some inflow from seeps and springs has smaller dissolved-solids concentrations (Cordova and others, 1972, p. 54-55) compared with those at site 16.2 (table 8). Much of this spring and seep water enters the main channel directly and from small tributaries downstream from the irrigation diversion, or as return flow from irrigation between sites 16.2 and 16. These sources cause the river to regain a significant flow in relation to the quantity diverted. Dissolved-solids concentrations in the Virgin River increased 11 percent through the irrigated area.

There was a net increase in dissolved-solids concentration between sites 29 and 16 during all of the sampling periods except February 1982 (fig. 4). Sodium hazard decreased from high to medium and salinity hazard remained very high between the two sites in August 1981 when flow was low (fig. 2). Little change occurred in May 1982 when flow was high (fig. 3). Water at site 16 was a sodium calcium sulfate chloride type during low flow and a calcium sodium sulfate bicarbonate chloride type during high flow (table 7).

Fort Pierce Wash

About one-third of the Virgin River drainage is in Arizona. Water from most of this area reaches the river through Fort Pierce Wash. Virtually all land is uninhabited desert and is used for limited grazing. Altitudes generally range from 2,500 to 5,000 feet except for Mt. Trumbull (altitude 8,028 feet) at the southern edge and an area west of Hurricane fault near the Utah-Arizona border. Gradients are irrelevant along most of the channel because the wash is mainly ephemeral and flows that do occur are discontinuous. Gradient near the mouth of the channel is less than 1 foot per mile.

A subdivision and golf course presently span the channel near the mouth (fig. 6). Development along the channel was noted in a flood-plain study of the Virgin River and Fort Pierce Wash area (U.S. Department of Defense, 1973, p. 3).

Two main tributaries to the wash are Short Creek, which originates in the eastern part of the drainage and flows generally westward along the Utah-Arizona border, and Hurricane Wash, which originates in the southern part of the drainage near Mt. Trumbull and flows north along the west side of Hurricane fault. These streams are ephemeral except for a small reach of Short Creek upstream from Colorado City. This water is diverted and used for limited irrigation in that area.

Flow in Fort Pierce Wash upstream from the farmed area south of St. George comes from flash floods and runoff during the spring season. A water sample collected for this area (site 15.2) in May 1982 contained large concentrations of calcium and sulfate. Seepage and return flow from irrigation through the farmed area contribute little to flow in the wash. Dissolved-solids concentrations of the water at site 15 were similar to those of the river water at site 6 except during May 1982, when they were double those of the river.

Santa Clara River

The Santa Clara River originates on the north side of Pine Valley Mountains, flows west and then south and southeast to join the Virgin River south of St. George. Altitudes along the Santa Clara River range from about 8,000 feet in the headwaters to 2,500 feet at the mouth, with most of the altitude change being between the headwaters and Gunlock. The upstream reach has a gradient of about 175 feet per mile mostly through igneous rocks and is in a narrow canyon with little vegetation along the banks. By contrast, the downstream reach from Gunlock to the Virgin River has a gradient of about 52 feet per mile, is in a wider canyon, has medium growth of trees and willows, and flows mostly through fine- and course-grained clastic sedimentary rocks.

Water samples collected at sites 14, 14.1, and 14.2 had larger dissolvedsolids concentrations in August 1981, when the flow was small, than in May 1982 when the flow was about 20 times larger than in August 1981 (table 4). Dissolved-solids concentrations decreased progressively downstream in August 1981, indicating that tributary inflow was diluting dissloved-solids load. The opposite occurred in May 1982, although little overall change was noted. Flow, with less dissolved solids, from springs entering the main channel

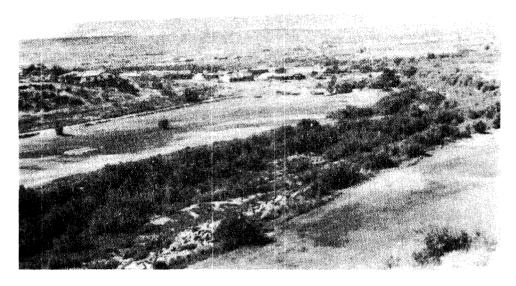


Figure 6.—Fort Pierce Wash near mouth with golf course and housing subdivisions. Site 15 is at far end of the golf course.

between sites 14.2 and 14 probably was the reason for a decrease in dissolvedsolids concentations downstream in August 1981. This flow was minor and probably had a negligible effect when the streamflow was high in May 1982.

Flow usually occurs only during snowmelt. Dissolved-solids concentration of the 1 sample collected was 58 milligrams per liter, similar to that of other samples along the river upstream from site 13.3. A spring at site 13.3 flowing about one-half as much as low river flow and having more than twice the dissolved-solids concentrations, significantly increased mineral content of the river downstream from this point.

Veyo Hot Springs (site 12.1), about 6 miles upstream from Gunlock, flows from the base of a nearly vertical basalt canyon wall. Water temperature was 31 and 16 ^OCelsius and flow was 0.67 and 0.75 cubic foot per second for two samples collected. Mundorff (1970, p. 43, 44) reported a temperature of 98 ^OFahrenheit or 37 ^OCelsius and a flow of 120 gallons per minute or 0.27 cubic foot per second and indicated that water may be of meteoric origin. Dissolved-solids concentration was about 25 percent greater than in the river water but the spring had virtually no effect on the river quality because of the relatively small spring inflow. Water from these springs is used in a swimming pool before entering the stream.

The Santa Clara River is diverted to a small reservoir downstream from Veyo Hot Springs for generating power and re-enters the main channel at Gunlock where it flows into Gunlock Reservoir. Several tributaries enter the main channel between the reservoir diversion and Gunlock, but little water flows as far as Gunlock except during periods of snowmelt or flash flooding. Samples were collected near the mouth of two perennial triburaries, Moody Wash (site 10) and Matgotsu Creek (site 11). Major ions in both samples were calcium and bicarbonate.

Water from Gunlock Reservoir is used for irrigation prior to entering the Virgin River. Use and return flow of this water between sites 8 and 7, mouth of the Santa Clara River, increased the dissolved-solids concentration about 500 percent. Largest ion increases were calcium, sulfate, and boron.

Virgin River Drainage Area from Santa Clara River to Littlefield, Arizona

Altitudes along this reach of the Virgin River range from about 2,500 feet at site 6 to 1,840 feet at site 1, with about two-thirds of the change occurring in Virgin River Gorge located between the Utah-Arizona border and Littlefield, Arizona. Reaches at either end of the gorge are meandering sandy channels with considerable vegetation along the banks. The channel through the gorge is narrow and rocky with little vegetation. Many dry streambeds join the main channel along the entire reach. Inflow to the river from Fort Pierce Wash at site 15 and Santa Clara River at site 7 affected the quality of the Virgin River at site 6 during August and October 1981, and February 1982, but had little affect during May and August 1982 (fig. 4). Variations in changes were caused by combination of flows and mineral concentrations. Sodium and salinity hazards were virtually unchanged by the inflow (figs. 2 and 3). Major ions at site 16 and 6 were sodium, calcium, sulfate, and chloride during low flow. Major ions during high flow were the same with the addition of bicarbonate.

The entire river flow seeps into the streambed about 6 miles downstream from site 6 during extremely low flow, and the estimated average annual loss along this reach is 50 cubic feet per second (U.S. Bureau of Reclamation, 1982, p. 12). Geological Survey personnel measured losses of 36 to 106 cubic feet per second in 1952 and 1956. The streambed is thence dry to Virgin River Gorge where about 50 springs discharge into the river, with an additional 20 springs discharging into the river between the gorge and Beaver Dam Wash (Trudeau, 1979, pl. 3). Flow ranges from 0.1 to 3.0 cubic feet per second for individual springs, with a composit flow of about 60 cubic feet per second. Tritium analyses indicate at least two different sources of recharge to the springs; one was a minimum of 22 years old and one less than 22 years old. These sources are probably river seepage from upstream and local recharge from precipitation (U.S. Bureau of Reclamation, 1982, p. 12). Trudeau (1979, p. 50, 51) reported dissolved-solids concentrations in the springs of 2,940 milligrams per liter, with a predominance of calcium and sulfate. Bicarbonate and sodium, concentrations also were large. Temperature ranged from 22 to 27 ^OCelsius.

Dissolved-solids concentrations in the Virgin River decreased between sites 6 and 3 during the August 1981 sample period when the riverbed was dry along part of the reach but increased during other periods when flow between sites was continuous (fig. 4). Concentrations of all constituents increased during later periods, with the largest increases occurring in calcium, sodium, sulfate, and chloride. The section of dry riverbed was mainly through geologic formations containing significant quantities of these constituents.

Beaver Dam Wash enters the river from the northwest, 1 mile upstream from Littlefield. Headwaters are located about 50 miles north of Littlefield along the Utah-Nevada border. The stream is mostly in Utah and flows in a southerly direction. It is ephemeral except for short perennial reaches. Five sites (2 to 2.4) were sampled during the study. Reaches were discontinuous and waters were not necessarily related. All flows upstream from site 2 ranged from 2 to 5.3 cubic feet per second and contained from 200 to 500 milligrams per liter of dissolved solids. Major ions were calcium and bicarbonate. Springs near the mouth of the wash (site 2) supply the only water to the Virgin River. Flows during all sample periods were less than 6 cubic feet per second and contained less than 700 milligrams per liter of dissolved solids. Major ions at site 2 were calcium, sodium, magnesium bicarbonate, sulfate, carbonate, and bicarbonate. Little effect on water quality in the Virgin occurred by the inflow from Beaver Dam Wash.

Flow in the Virgin River increased a maximum of 9 percent between Beaver Dam Wash and Littlefield (site 1) as a result of springs and seeps entering the river. This flow, in addition to inflow from Beaver Dam Wash, caused no change in dissolved solids of the river in May 1982 between sites 3 and 1 and slight changes during other sampling periods (fig. 4). Sodium hazard was low and salinity hazard was very high in August 1981 at site 1. Water was a calcium sodium magnesium sulfate chloride type. Long-term records at site 1 include 609 analyses. Calcium and sulfate were ions with the greatest mean concentration for the period of record (1949-78) although other ions may have been predominant at specific times (table 9). Potassium and nitrogen were ions with the least mean concentration.

Trace Elements

Constituents of natural waters typically occurring in concentrations of less than 1.0 milligram per liter commonly are referred to as trace elements (Hem, 1970, p. 188). Water samples for trace-element analysis were collected at 23 sites in the Virgin River basin August 18-31, 1981. These samples were analyzed semiquantitatively for 24 constituents (table 2).¹ Based on those analyses six sites were resampled from August 23 to September 3, 1982 for quantitative analyses of 10 trace elements (table 7). The 10 trace elements analyzed for quantitatively were chosen based on previous reports of possible problems in the basin and constraints imposed by available laboratory schedules.

The Five County Association of Governments (1977) inventory of Virgin River basin sites reported sporadic problems with large concentrations of the elements arsenic, iron, manganese, and selenium. Semiguantitative and quantitative analyses for this study (table 7) showed iron or selenium concentrations less than the permissable minimum contaminant level of the U.S. Environmental Protection Agency (1972) although reported concentrations are dissolved only. One quantitative analysis (site 21, September 2, 1982) showed an arsenic concentration larger than the Utah State Division of Health (Utah Water Research Laboratory, 1974) recommended limits in the dissolved phase; but the concentration was lower than the mandatory limit. Samples from six sites in the Virgin River and tributaries had dissolved manganese concentrations larger than recommended Utah Class C stream standard concentrations and in excess of U.S. Environmental Protection Agency (1972) public-water supply and irrigation recommended limits for total manganese concentration.

Other Factors Affecting Surface-Water Ouality

Fluvial Sediment

Sediment concentration depends on runoff intensity and type, and stability of watershed material. Concentration and quantity of runoff, or discharge, determines total sediment load of a stream. Relation between suspended-sediment concentration and load is expressed by the formula:

Suspended-sediment load = discharge x concentration x 0.0027.

Suspended-sediment load is reported in tons per day, discharge in cubic feet per second, concentration in milligrams per liter, and 0.0027 is the unit's conversion factor.

Suspended sediment in streams of Virgin River drainage mostly is transported during thunderstorms and snowmelt. Suspended-sediment loads caused by thunderstorms vary in intensity and pattern at different locations. Thunderstorms are limited in extent and usually affect only parts of the basin where they occur and the channel downstream. Loads during snowmelt may show similar patterns in larger parts of the basin.

¹Lead concentrations are not reported in the semiquantitative analysis table 7 because of an error in the methodology.

Suspended-sediment analyses for this study represent only concentration and loads during stable flow conditions as sampling periods were selected when streamflow was most stable. Suspended-sediment samples were collected at 25 sites in August when flows were low, and at 19 sites in May when flows generally were high. The largest load recorded during the study was 34,108 tons per day during a flood on August 24, 1982 at site 17. Sediment loads ranged from 0.06 (site 33) to 2,555 tons (site 16.2) per day in August and from 0.55 (site 77) to 3,580 tons (site 33.1) per day in May. Comparison of data for the two periods shows the greater transporting capacity of the larger streams in May. Stream discharge and suspended-sediment loads at selected sites for these periods are shown in table 8.

Sediment data were collected at six sites within the study area for various long-term periods. These data show a large variation in sediment load of the river and its tributaries. Maximum and minimum loads for periods of record at these sites are shown in table 3.

Suspended-sediment discharge during water year 1970 at site 32 is shown in table 4. During the 1970 water year total annual sediment load was near average for the 9-year period of record at site 32. It is, therefore, considered to be representative of annual sediment loads at the site.

Rapid velocities resulting from steeper gradients and large discharges increases the erosive power in upstream reaches of streams and produce larger sediment loads per square mile. The following table compares altitude changes, drainage areas, and sediment loads in upstream and downstream reaches of the Virgin River; the 1963-68 period was selected because it was the only period common to site 32 and 1:

	Stream	Drainage	Sediment load				
Site gra no. (fe	Stream gradient	Drainage area	nage 1963-68 averag		Ма	Maximum	
	(feet per mile)		Tons per day	Tons per day per square mile	Tons per day	Tons per day per square mile	
54							
32	121	934	3,870	4.1	1,300,000	1,392	
1	30	4,156	5,929	1.4	1,800,000	433	

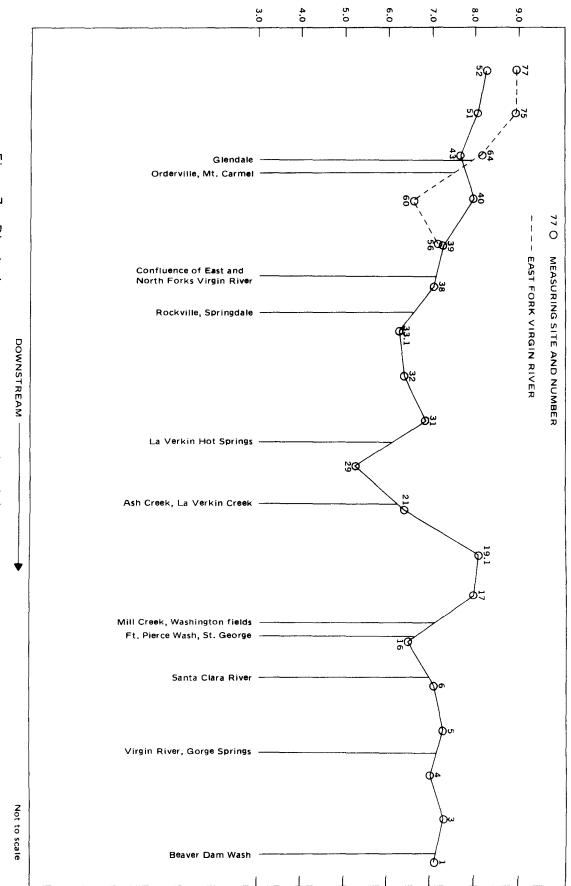
Dissolved Oxygen

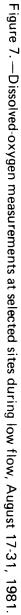
Determination of dissolved-oxygen concentration was made during each site visit (table 4). Concentrations, excluding those at hot springs, ranged from 5.2 milligrams per liter (79-percent saturation) at site 29 in August 1981, to 12.5 milligrams per liter (105-percent saturation) at site 56 in February 1982. Dissolved-oxygen concentrations of 0.5 and 4.6 milligrams per liter were noted at the two hot springs in August 1981; however, after mixing with receiving water, the effect of the smaller dissolved-oxygen concentrations was largely negated. Dissolved-oxygen concentration at site 29, immediately downstream from La Verkin Hot Springs, was 5.2 milligrams per liter on August 25, 1981 during low flow. This is near the minimum concentration of 5.0 milligrams per liter recommended by the U.S. Environmental Protection Agency (1976b, p. 123) for maintenance of a viable fish population and is less than Utah Class C stream standards that have been applied to the Virgin River system (Utah Water Research Laboratory, 1974, p. 15). When combined with increased temperatures, this may create an avoidance barrier to fish during low flows (Warren, 1971, p. 188) and affect distribution and survival of aquatic insects (Nebeker, 1972). Dissolved-oxygen concentrations at selected sites in August 1981 are shown in figure 7.

Diurnal dissolved-oxygen periodicity was measured at 11 sites with a continuous dissolved-oxygen monitor for varying times during March and April 1982. Graphic representation of dissolved-oxygen fluctuations, combined with separate site measurements, showed that dissolved-oxygen concentrations in the basin correlated closely with water temperatures in all areas not directly affected by thermal springs. Monitor data on some tributaries showed marked increases in dissolved-oxygen concentrations during daylight hours due to photosynthetic oxygen production. Turbid conditions and the shifting nature of the sand channel in the mainstream limit aquatic vascular plant, phytoplankton, and algal growth so that the photosynthetic contribution to dissolved-oxygen concentrations is minimal. At monitoring sites on the Virgin River there was almost a direct correlation between water temperature and dissolved-oxygen concentrations without large photosynthetic daytime peaks.

Continuous monitoring downstream from La Verkin Hot Springs (site 29), February 26 to March 3, 1982, indicated an anomaly for both dissolved oxygen and water temperature. For about 1 hour each evening between 7:30 and 11:30 p.m. (local time), the dissolved-oxygen concentrations were rapidly decreased by 0.4 to 0.6 milligram per liter and after 1 hour rapidly increased to previous concentrations. A corresponding increase in water temperature of 1.3 to 1.6 ^OCelsius was noted during each period of dissolved-oxygen decrease. An increase in specific conductance of the water also was noted on a separate monitor; but as this was an hourly monitor, it did not consistently register the increase. A similar irregularity of lesser intensity was noted during the morning hours.

Specific conductance, pH, dissolved oxygen, and water temperature were monitored at the springs during the evening of October 28, 1982. Although the hot springs pool water was observed to become very milky during the monitoring, no significant changes were noted in any of the measured properties. The oxygen depletion monitored downstream from the springs was, therefore, attributed to a brief increase in spring flow in a consistent geyser-like fashion and to incomplete mixing with surface water at the





67 DISSOLVED-OXYGEN, IN MILLIGRAMS PER LITER monitoring site, as noted by measurements made at cross sections on September 22, 1982. Further study will be needed to confirm the exact cause and effect relationship.

Bacteria

Certain bacteria which inhabit the intestines of animals are widely used in detecting fecal contamination and the possible presence of pathogenic organisms. While the "indicator bacteria" themselves may not be harmful, they are considered to indicate the disease-producing potential of water. Fecal coliform bacteria are among the organisms present in the intestine and feces of warm-blooded animals and are used to indicate recent contamination. Fecal streptococcus bacteria also are present in the intestine of man and other animals and may aid in determining the origin of pollution (Slack, 1974, written commun.).

Bacterial samples were collected at 24 sites during August 18-31, 1981. Samples were analyzed using the membrane-filter method (Greeson and others, 1977). Results of the bacterial sampling are presented in table 5.

Also included in table 5 is the ratio formulated by dividing fecal coliform colony (FC) count by the fecal streptococcus colony (SC) count (both expressed as colonies per 100 milliliters of sample). The FC/FS ratio has been used as an investigation tool in pinpointing the source and location of fecal-waste contamination (American Public Health Association and others, 1976).

Determination of the origin of fecal contamination is based on the following values for the ratio (Millipore Corp., 1972, p. 36):

Fecal coliform count (colonies per 100 milliliters) = FC/FS ratio Fecal streptococcus count (colonies per 100 milliliters) FC/FS24.0 indicates pollution derived from human wastes.

0.7 < FC/FS<u>≤</u>4.0 indicates mixed pollution sources. FC/FS<u>≤</u>0.7 indicates pollution primarily from livestock or poultry sources.

The U.S. Environmental Protecton Agency (1972, p. 58) recommends that fecal coliform densities in untreated surface-water sources not exceed 2,000 colonies per 100 milliliters.

Data obtained from the single sampling are not sufficient to confirm sanitary problems in the basin; however, earlier studies have documented large coliform bacteria concentrations in tributaries and in the main channel during low flow (Five County Association of Governments, 1977, p. 61).

Pesticides in Stream-bottom Materials

Analyses were made for 24 organochlorine or organophosphorus compounds in bottom-material samples at 4 sites downstream from major agricultural areas. Results of the pesticide analyses are presented in table 6. None of the compounds were present in detectable concentrations.

SUMMARY

Water quality in the upper Virgin River basin from the headwaters in Utah to Littlefield, Arizona, varies considerably. This variation results primarily from variations in geology, hot-spring inflow, and return flow from irrigation. During this study, dissolved-solids concentrations were less than 100 milligrams per liter in the headwaters of North Fork Virgin River and the Santa Clara River. Headwaters of other triburaries generally contained between 200 and 400 milligrams per liter dissolved solids. Dissolved-solids concentrations increase progressively downstream in each tributary, which increased or decreased concentrations in the receiving stream depending on relative discharges of the two streams.

Concentrations of dissolved solids in the Virgin River were large during low flow when inflow from tributaries had greatest effect on the river. Many small tributaries had larger concentrations than the river at higher flows but had little effect on the river quality because their flow was small compared to that of the river.

La Verkin Hot Springs caused the single greatest change in river quality. In the Virgin River downstream from La Verkin Hot Springs sodium hazard generally was low, but was medium to high during low flow. Salinity hazard ranged from low to high upstream from the springs and high to very high downstream from the springs. Dissolved-solids concentration downstream from the springs was nearly five times greater than upstream from the springs during the lowest flow and less than two times greater during the highest flow. Boron concentrations upstream from the spring were smaller than the tolerance level of all boron-sensitive crops; they were larger than tolerance level of many crops downstream from the springs.

Calcium was the predominant cation and sulfate the predominant anion in the river system. Sodium, magnesium, bicarbonate, and chloride were in greater concentrations in some reaches, but not as consistently present as calcium and sulfate.

Most sediment was transported by the river during thunderstorms and snowmelt. Greatest suspended-sediment loads were measured in the southwestern part of the basin with the greatest loads per square mile occurring in the northeastern part of the upper basin.

Dissolved-oxygen concentrations generally were larger than 6.0 milligrams per liter within the basin except in short reaches immediately downstream from thermal springs and in several small tributaries during low summer flows.

Manganese concentrations were in excess of recommended limits in some reaches of the river and in several small tributaries.

Data from this study were insufficient to confirm sanitary problems. These data, combined with previous studies, indicate that bacterial loads may cause degraded sanitary conditons downstream from livestock grazing areas during low flow. Pesticide concentrations were minimal at all sites sampled.

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Table 1.--Mixing characteristics of La Verkin Hot Springs and Virgin River waters

Section 1: located about 300 feet downstream from main spring area. Section 2: located about 600 feet downstream from main spring area.

Asterisk (*) indicates small spring located on right bank about 5 feet upstream from section 1.

		Section 1		<u></u>	Section 2	
Distance from bank (feet)	Temperature (degress Celsius)	Specific conductance (microhmos per centimeter at 25 ° Celsius)	Dissolved oxygen (milligrams per liter)	Temperature (degrees Celsius)	Specific conductance (microhmos per second at 25 ⁰ Celsius)	Dissolved oxygen (milligrams per liter)
(left bank)						
0	36	15200	3.2	26	5100	5.7
3	29	6500	4.8	26	5000	5.7
6	26	4100	5.2	26	5000	4.6
9	25	3650	6.2	25	490 0	5.7
12	24	3500	6.4	24	4200	6.0
15	24	3400	6.4	24	3700	6.3
18	24	3300	6.5	24	3600	6.4
21	24	3300	6.5	23	3600	6.4
24	24	3300	6.5	23	3700	6.4
27	24	3400	6.5	22.5	3700	6.4
30	24.5	3470	6.4	23.5	3780	6.4
33 36	24.5	3550 3600	6.3	24	3800	6.4
39	24.5 25		6.2 6.2	24 22 F	3880	6.2
42	25	3 800 5100	5.2	23.5 22	3900 3900	6.2 6.1
(right bank))					
*	38	15500	0.2			

			<u></u>										
Site No.	Sampling Date	Site Name	Alu- minum, (Al)	Arse- nic, (As)	Anti- mony, (Sb)	Bar- ium, (Ba)	Beryl- lium, (Be)	Bi s- muth, (Bi)	Cad- mium, (Cd)	Chro- mium, (Cr)	Co- balt, (Co)	Cop- per, (Cu)	Gal- lium, (Ga)
2	8-28-81 ¹	Beaver Dam Wash at mouth, Arizona	100		<30	100	<1	<1,000	<1	<50	<5	<10	<30
4	do.	Virgin River bel ow Virg in River Gorge, Utah	700		50	30	<1	<1,000	<1	<50	<5	<10	100
	9- 3-82	do.		8			-		<1	20		1	
6	8 -28-81	Virgin River near Bloomington, Utah	700		50	100	a	<1,000	1	<50	<5	<10	100
7	8-18-81	Santa Clara River at mouth, Utah	500		50	70	<1	<1,000	<1	<50	<5	<10	50
	8-23-82	do.	-	8					<1	20		1	
12	8-18-81	Santa Clara River below Veyo Hot Springs, Utah	300	-	30	100	4	<1,000	<1	<50	<5	<10	<30
12.1	do.	Veyo Hot Springs, Utah	100		<30	100	<1	<1,000	<1	<50	<5	<10	30
13	do.	Santa Clara River above Veyo Hot Springs, Utah	100		30	100	d	<1,000	<1	<50	<5	<10	<30
15	8-28-81	Fort Pierce Wash at mouth, Utah	500		<30	100	4	<1,000	<1	<50	<5	<10	100
	9-2-82	do.	-	10					<1	20		2	
16.1	8-20-81	Mill Creek near Washington, Utah	100	-	<30	50	<1	<1,000	<1	<50	<5	<10	<30
18	8-19-81	Leeds Creek at mouth, Utah	500		70	100	<1	<1,000	<1	<50	<5	<10	<30
19.1	8-31-8 1	Virgin River above Leeds Creek, Utah	300	-	30	70	<1	<1,000	<1	<50	<5	<10	50
20.1	8-28-81	Gould Wash at mouth, Utah	300		<30	100	<1	<1,000	<1	<50	<5	<10	<30
21	8-25 -61	Virgin River below Ash Creek, Utah	500	_	50	100	ব	<1,000	1	<50	<5	<10	100
	9- 2-82	do. '	-	31			~	_	<1	10		<1	
23	8-25-81	Ash Creek below Westfield Ditch, Utah	300	-	30	70	<1	<1,000	1	<50	<5	<10	50
	9 -2-82	do.	-	4		-			<1	<10		<1	
27	8-25-81	La Verkin Creek at mouth, Utah	300	-	30	100	<1	<1,000	<1	<50	<5	<10	50
	9- 2-82	do.	-	5				·	<1	<10		1	
30	ት-25- 80	La Verkin Hot Springs, Utah	700		70	50	<1	<1,000	7	<50	<5	<10	100
32	8-24-81	Virgin River at Virgin, Utah	300	—	<30	100	<1	<1,000	1	<50	<5	\10	70
33	do.	North Creek at mouth, Utah	300	_	30	100	<1	<1,000	<1	<50	<5	<10	30
39	do.	North Fork Virgin River above confluence, Utah	100		30	100	<1	<1,000	<1	<50	<5	- 10	<30
40	do.	North Fork Virgin River near Springdale, Utah	100		30	100	<1	<1,000	<1	<50	<5	<10	<30
56	do.	East Fork Virgin River above confluence, Utah	100		<30	100	<1	<1,000	<1	<50	<5	<10	<30
60	8-20 -81	East Fork Virgin River at Mt. Carmel, Utah	500	-	<30	70	<1	<1,000	<1	<50	<5	<10	<30
64	do.	East Fork Virgin River near Glendale, Utah	300		<30	50	<1	<1,000	<1	<50	<5	<10	<30

1 1981 data are semiguantitative. Results are reported from detection limit to upper concentration limit in steps of 1, 3, 5, 7, and 10. Due to rounding technique results are an estimate of one significant figure. Precision is approximately (+) or (-) one step at 68 percent confidence level and (+) or (-) two steps at 95 percent confidence level.

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•		м	icrograms	per liter											
Germa- nium, (Ge)	Iron, (Fe)	Lead, (Pb)	Lith- ium, (Li)	Manga- nese, (Min)	Mer- cury, (Hg)	Molyb- denum, (Mo)	Nickel, (Ni)	Sele- nium, (Se)	Sil- ver, (Ag)	Stron- tium, (Sr)	Tin, (Sn)	Tita- nium, (Ti)	Van a dium, (V)	Zinc, (Zn)	Zirco- nium, (Zr)
300	<5	_	50	1		<10	<50		<10	1,000	100	<5	30	<5	<5
700	10		500	70		10	<50		<10	5,000	700	10	<10	10	<5
	30	4	_	110	0.1		_	2						10	
700	10		500	100		10	<50		10	5,000	700	10	30	<5	<5
500	10	_	70	300	<u></u>	<10	<50		<10	3,000	500	7	10	<5	<5
	11	5		420	<.1			2	·	·				11	
100	7	-	30	7		<10	<50		<10	300	100	<5	<10	<5	<5
300	5		30	1		<10	<50		<10	700	300	<5	10	<5	<5
100	7		30	7		<10	<50		<10	300	100	<5	10	<5	<5
700	10		300	300		<10	<50		<10	5,000	500	10	<10	7	<5
	20	<1		120	.1			1						20	
100	7		· 100	10		<10	<50		<10	1,000	100	<5	<10	<5	<5
700	10		70	10		<10	<50	_	<10	3,000	500	<5	10	<5	<5
500	7		500	5	_	<10	<50		<10	3,000	300	7	<10	<5	<5
300	30		30	5	_	<10	<50		<10	1,000	100	7	<10	<5	<5
500	10	_	500	50		10	<50		10	3,000	500	10	<10	10	5
	20	5		70	<.1			1						10	
300	10		10	3		10	<50	_	<10	1,000	300	10	10	<5	<5
	<3	<1		2	.1			1						3	
500	10		30	100		<10	<50		<10	3,000	300	10	10	<5	<5
	8	<1		210	.1			2						20	
1,000	10		3,000	30		30	<50	-	<10	>10,000	700	30	10	10	<5
300	10		30	7		10	່ <50		10	1,000	300	10	<10	<5	<5
300	10		50	30		<10	<50		<10	3,000	300	7	10	10	<5
300	7		30	5		<10	<50		<10	700	100	<5	<10	<5	<5
100	<5		30	10	-	<10	<50	·	<10	700	100	<5	<10	<5	<5
300	<5	_	30	7		<10	<50		<10	1,000	100	<5	<10	5	<5
500	5		70	10		<10	<50	_	<10	1,000	500	<5	<10	<5	<5
300	30		30	30		<10	<50		<10	300	300	<5	<10	<5	<5

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collected during August 1981 and August, September 1982

Table 3.--Recorded maximum and minimum suspended-sediment loads at six sites

in the Virgin River drainage

			Maximum suspended sediment				suspended diment	
				Mean			Mean	
			đ	ischarge			discharge	
	Gaging	Period		(cubic			(cubic	
Site	station	of		feet per	Load		feet per	Load
no.	number	record	Date	second)	(tons/day)	Date	second)	(tons/day)
1	09415000	Oct. 1947 to Sept. 1968 ¹	12- 7-66	15,200	1,800,000	6-16-62	58	6
						6-20-62	55	
k	09410000	May 1962 to Sept. 1968	9- 28-62	359	540,000	many days		<.5
17	09408150	Mar. 1967 to June 1974	9– 19–72	4,340	1,930,000	6-19-74	57	2.9
32	09406000	May 1962 to June 1971	12- 6-66	9,670	1,300,000	7-11 - 69	94	4.6
51	0 940 54 50	Oct. 1977 to Sept. 1982	9-10-80	49	814	9- 8-81	7.7	.07
52	09405420	Oct. 1974 to Sept. 1977	10- 2-76	31	1,080	10-20-75	6.6	.02

1 Also monthly suspended-sediment records for 1978 water year.

Asterisk (*) indicates discontinued gaging station located between sites 8 and 8.1 prior to construction of Gunlock Reservoir.

Table 4.---Suspended sediment at site 57, "From Piver at Virgin. U.s. (Water year October 1969 to September 1970)

		October			November			Decemberi	
-		Mean			Mean	· · · · · · · · · · · · · · · · · · ·	······································	Mean	
	Mean	concen-	Sediment	Mean	concen-	Sediment.	Mean	concen-	Sediment
Day	discharge	tration	discharge	discharge	tration	discharge	discharge	tration	dischar9€
	(cubic feet per second)	(milligrams per liter)	(tons per day)	(cubic feet per second)	(milligrams per liter)	(tons per day)	(cubic feet per second)	(milligrams per liter)	(tons per dav)
	per second	per (itter)		per second,				pet ricery	
1	82	1000	221	144	999	3.88	150	812	329
2	86	720	167	141	856	326	153	949	3.67
3	90	51.8	126	139	734	275	156	1110	468
4	92	518	129	135	600	219	161	1020	443
5	101	518	141	136	600	220	158	93.8	400
6	1.05	916	260	132	600	214	14.9	862	347
7	101	748	204	187	1100	555	1.52	792	325
8	101	611	167	174	1600	752	156	394	166
9	101	666	182	163	946	416	163	196	86
10	101	725	198	167	650	293	156	433	182
11	105	790	224	165	242	108	147	957	3 80
12	108	861	251	155	502	210	152	713	2 93
13	100	832	225	152	1040	427	1.46	531	209
14	103	804	224	148	2160	863	148	396	158
15	105	777	220	152	2060	845	148	295	118
16	108	751	219	259	4370	3060	148	408	163
17	110	751	223	342	3500	3230	153	564	233
18	120	751	243	257	1300	90 2	153	779	322
19	140	740	280	152	1090	447	152	718	2.95
20	135	729	266	168	915	415	161	66 t	26.9
2.1	140	730	276	175	768	363	150	609	2.47
22	162	1850	80.9	171	645	298	155	512	214
23	154	1550	644	.164	541	240	154	430	179
24	149	1130	455	159	454	195	149	361	145
25	141	700	266	156	223	94	148	303	121
26	144	413	161	157	233	99	151	255	104
27	138	900	335	152	243	100	151	214	87
28	136	900	330	145	253	99	130	231	81
29	139	708	266	143	264	102	118	249	79
30	140	558	211	142	200	77	121	322	105
31	145	439	172		-		122	416	137
Total	3682		8095	5032		15832	4599		7072

	January				February		March			
Day	Mean discharge (cubic feet per second)	Mean concen- tration (milligrams per liter)	Sediment discharge (tons per day)	Mean discharge (cubic feet per second)	Mean concer- tration (milligrams per liter)	Sediment discharge (tons per day)	Mean discharge (cubic feet per second)	Mean concen- tration (milligrams per liter)	Sediment discharge (tons per day)	
1	124	350	117	132	300	107	312	6 800	5730	
2	116	32.8	103	127	272	93	4 80	10400	13500	
3	102	357	98	124	247	83	416	6100	6 850	
4	108	318	93	137	223	82	405	1040	1140	
5	110	283	84	134	201	73	403	869	946	
6	102	252	69	132	181	65	271	726	531	
7	114	225	69	133	230	83	156	575	242	
8	116	200	63	133	292	105	157	455	193	
9	130	269	94	128	406	140	148	46 9	187	
10	144	362	141	128	192	. 66	156	4 83	203	
11	143	4 87	188	135	242	88	151	4 97	203	
12	140	655	248	149	305	123	138	492	183	
13	138	655	244	144	403	157	134	488	177	
14	133	654	235	134	442	160	143	4 83	186	
15	155	654	274	127	4 84	166	151	530	216	
16	150	600	243	121	399	130	143	5 83	224	
17	186	800	402	121	329	107	146	478	188	
18	163	590	260	117	271	86	134	3.93	142	
19	144	661	257	118	223	71	122	323	106	
20	142	740	284	114	200	62	126	266	90	
21	143	829	320	125	200	68	125	2 93	99	
22	137	765	283	191	2500	1290	120	323	105	
23	147	706	280	165	1500	668	124	328	110	
2.4	150	651	264	135	81.4	297	124	332	111	
25	145	601	235	129	442	154	136	337	124	
26	137	601	222	129	240	84	131	342	121	
27	136	601	221	131	310	110	127	293	100	
28	137	2 37	106	130	400	140	119	251	<u>81</u>	
29	124	137	46				124	195	65	
30	119	219	70				129	151	53	
31	131	351	124				120	150	49	
Total	4166		5737	3723		4858	5671		32255	

		April			May			June	
- Day	Mean discharge (cubic feet per second)	Mean concen- tration (milligrams per liter)	Sediment discharge (tons per day)	Mean discharge (cubic feet per second)	Mean conCen- tration (milligrams per liter)	Sediment discharge (tons per day)	Mean discharge (cubic feet per second)	Mean concen- tration (milligrams per liter)	Sediment discharge (tons per day)
1	111	150	45	116	250	78	79	150	32
2	109	181	53	119	356	114	78	265	56
3	116	176	55	139	167	63	79	40	8.5
4	114	172	53	167	78	35	76	91	19
5	116	248	78	182	242	119	80	205	44
6	126	358	122	182	750	369	95	200	51
7	148	516	206	205	2100	1160	147	3670	1460
8	143	5.87	227	177	1800	860	89	2200	52 9
9	138	668	249	191	2100	1080	91	2400	590
10	151	761	310	209	2400	1350	98	2650	701
11	166	866	388	238	2100	1350	103	16 80	467
12	143	344	133	198	769	411	98	894	237
13	119	191	61	178	1280	615	89	781	188
14	120	133	43	190	1050	539	86	682	158
15	107	92	27	179	856	414	83	813	182
16	102	74	20	192	700	363	79	416	89
17	102	59	16	196	2600	1380	75	600	122
18	102	47	13	181	1240	606	76	337	69
19	103	67	19	157	592	251	74	360	72
20	103	96	27	136	333	122	74	384	77
21	97	75	20	119	292	94	84	771	175
22	101	59	16	106	248	71	83	758	170
23	98	48	13	106	258	74	83	391	88
24	92	81	20	103	375	104	81	3 85	84
25	97	136	36	107	257	74	82	379	84
26	112	230	70	93	143	36	81	373	82
27	184	286	142	90	212	52	85	556	128
28	167	355	160	84	61	14	92	594	148
29	133	249	89	83	49	11	88	635	151
30	121	175	57	89	39	9.4	90	678	165
31	-			84	118	27			
Potal	3641		2768	4596		11845.4	2598		6426.5
		July			August			September	
Day	Mean discharge (cubic feet per second)	Mean concen- tration (milligrams per liter)	Sediment discharge (tons per day)	Mean discharge (cubic feet per second)	Mean concen- tration (milligrams per liter)	Sediment discharge (tons per day)	Mean discharge (cubic feet per second)	Mean concer- tration (milligrams per liter)	Sediment discharge (tons per day)
1	94	2500	635	82	3500	775	73	2520	497
2	94	2500	635	89	3,800	913	85	2050	470

Table 4 .--- Suspended sediment at site 32, Virgin River at Virgin, Utah--Continued

		July			August		September			
- Ъау	Mean discharge (cubic feet per second)	Mean concen- tration (milligrams per liter)	Sediment discharge (tons per day)	Mean discharge (cubic feet per second)	Mean concen- tration (milligrams per liter)	Sediment discharge (tons per day)	Mean discharge (cubic feet per second)	Mean concen- tration (milligrams per liter)	Sediment discharge (tons per day)	
1	94	2500	635	82	3500	775	73	2520	497	
2	94	2500	635	89	3 800	913	85	2050	470	
3	94	2500	635	102	4120	1130	87	2000	470	
4	119	6600	2120	236	3 8200	155000	100	2000	540	
5	124	1800	603	950	98000	251000	840	32000	72600	
6	131	3000	1060	520	120000	16 8000	350	13000	12300	
7	109	2600	765	240	55000	35600	150	6440	2610	
8	165	14700	6510	112	4790	1450	100	3190	861	
9	307	12000	11800	119	4500	1450	82	2350	520	
lõ	351	6930	6570	113	4230	1290	79	1730	369	
11	630	4000	6800	108	6020	1760	75	1440	292	
12	100	752	203	106	7030	2010	79	1200	256	
.3	79	756	161	125	82.00	2770	172	11000	5110	
4	77	700	146	127	9570	32 80	83	7000	1570	
.5	77	539	112	176	20000	9500	77	2900	603	
6	89	2540	610	450	33000	40100	75	1200	243	
7	92	12000	2980	150	19000	7700	78	1240	261	
8	125	36000	12200	650	50000	87 800	88	12 90	307	
9	112	17000	5140	500	65000	87800	88	1330	316	
0	114	42 90	1320	350	5000	4730	86	1040	241	
1	336	51300	46500	300	10000	81.00	85	808	185	
2	263	11000	7810	360	60000	58300	88	968	230	
3	102	10000	2750	150	5330	2160	92	1160	288	
4	204	113000	62200	92	3930	976	88	1130	268	
5	121	12100	3950	80	2 890	624	81	1090	238	
5	100	8720	2350	105	5000	1420	78	1060	223	
7	99	6280	16 80	130	14000	4910	79	978	209	
8	90	4820	1170	150	6220	2520	78	903	190	
9	82	3700	81.9	100	4610	1240	78	833	175	
)	88	3350	796	83	3420	766	80	800	173	
1	82	3040	673	74	2500	500				
al	4650		191703	6929		945574	3674		102615	
	ischarge for ye	ar (cubic feet nt discharge fo							52961 1334780	

	10010 37 11	ararcor a	ICCCLIG	adilpred hug.	18-28, 1981		
					Fecal	Fecal	(A)
				Discharge (cubic	coliform (FC) : (colonies	streptococci (F (colonies	S)
Site				feet per	per 100	per 100	FC/FS
no.	Site name	Date	Time	second)	milliliters)		Ratio
2	Beaver Dam Wash at mouth, Ariz.	8-28-81	1030	3.7	70	5800	0.01
4	Virgin River bel <i>o</i> w Virgin River Gorge Springs, Ariz.	8-28-81	1230	30.0	4 0	330	.12
6	Virgin River at Bloomington, Utah	8-28-81	1415	35.0	² 6200	9300	.67
7	Santa Clara River at mouth, Utah	8-18-81	1845	4.5	¹ , ² 2400	1 ₂₁₀₀	1,1
12	Santa Clara River below Veyo Hot Springs, Utah	8-18-81	1300	2.6	¹ , ² 2400	550	4.4
12.1	Veyo Hot Springs, Utah	8-18-81	1400	.67	1<1	$1_{<1}$	
14.1	Santa Clara River below con- fluence with Right Fork, Utah	8-18-81	1100	2.8	30	1 _{4 80}	.06
14.2	Santa Clara River above campground, Utah	8-18-81	0930	.86	90	300	.30
15	Fort Pierce Wash at mouth, Utah	8-28-81	1545	2.1	2 ₅₀₀₀	6000	.83
16.1	Mill Creek below Washington, Utah	8-20-81	1530	1.9	1,220,000	6500	3.1
18	Leeds Creek at mouth, Utah	8-19-81	1130	1.2	1>120	1200	
19.1	Virgin River above Leeds Creek, Utah	8-31-81	0945	69.0	500	1600	.31
20.1	Gould Wash at mouth, Utah	8-28-81	1800	1.4	² 26,000	44,000	.59
21	Virgin River below Ash Creek, Utah	8-25-81	1400	55.0	1200	1800	.67
23	Ash Creek below Westfield Ditch near Toquerville, Utah	8-25-81	0900	9.3	1 ₃₀	230	.13
27	La Verkin Creek at mouth, Utah	8-25-81	1100	2.2	1 ₅₀₀	1800	.28
30	La Verkin Hot Springs, Utah	8-25-81	1530	12.0	1 ₂₀	1 ₅	4.0
32	Virgin River at Virgin, Utah	8-24-81	1 800	80.0	11300	3700	.35
33	North Creek at mouth, Utah	8-24-81	1645	.85	11400	1300	1.1
39	North Fork Virgin River above confluence, Utah	8-24-81	1345	33.0	1500	1800	0.83
40	North Fork Virgin River near Springdale, Utah	8-24-81	1115	43.0	¹ 80	350	.23
56	East Fork Virgin River above confluence, Utah	8-24-81	1245	39.0	¹ 60	47 0	.13
60	East Fork Virgin River at Mt. Carmel Junction, Utah	8-20-81	1145	3.8	1 ₁₀₀₀	790	1.3
64	East Fork Virgin River near Glendale, Utah	8-20-81	1000	12.0	1>120	1>200	

Table 5.--Indicator bacteria sampled Aug. 18-28, 1981

¹ Based on colony count outside of the ideal range ² Exceeds criteria for untreated surface water intended for public-water supplies (U.S. Environmental Protection Agency, 1972)

₩,	Microgra	ams per kilogra	m (mg/kg)	
		Site name	<u></u>	<u>,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,</u>
Compound and insecticide	Santa Clara River at mouth, Utah (site 7)	Virgin River above Fort Pierœ Wash, Utah (site 16)	Virgin River above Leeds Creek, Utah (site 19.1)	East Fork Virgin River at Mt. Carmel Junction, Utah (site 60)
Aldrin	<0.1	<0.1	<0.1	<0.1
Chlordane	<1.0	<1.0	<1.0	<1.0
DDD	<.1	<.1	<.1	<.1
DDE	<.1	<.1	<.1	<.1
DDT	<.1	<.1	<.1	<.1
Diazinon	<.1	<.1	<.1	<.1
Dieldrin	<.1	<.1	<.1	<.1
Endosulfan	<.1	<.1	<.1	<.1
Endrin	<.1	<.1	<.1	<.1
Ethion	<.1	<.1	<.1	<.1
Gross PCB	<1.0	<1.0	<1.0	<1.0
Gross PON	<1.0	<1.0	<1.0	<1.0
Heptachlor Epoxide	<.1	<.1	<.1	<.1
Heptachlor	<.1	<.1	<.1	<.1
Lindane	<.1	<.1	<.1	<.1
Malathion	<.1	<.1	<.1	<.1
Methyl Trithion	<.1	<.1	<.1	<.1
Methyl Parathion Mirex	<.1 <.1	<.1 <.1	<.1 <.1	<.1 <.1
Methoxychlor	<.1 <.1	< . 1	<.1	<.1 <.1
Parathion	< . 1	< . 1	<.1	< . 1
Perthane	<.1	<.1	<.1	<.1
Toxaphene	<1.0	<1.0	<1.0	<1.0
Trithion	<.1	<.1	<.1	<.1

Table 6.--Organochlorine compounds organophosphorus insecticides, POB and PON in stream-bottom materials

[Abbreviations: ft³/s, cubic feet per second; umho, micromhos per centimeter

Site number: Numbers refer to sites shown on plate 1. Gaging station number: Standard Geological Survey gaging station number. The 09 indicates all drainage in the Colorado River Basin. The remaining Discharge range: Period of record for gaging stations, sampling periods for other sites. Dominant cations, anions: Refer to section "Classification of water for public supply and irrigation" for methods of determination. Ca, calcium; Mg, Irrigation supply: L-low, M-medium, H-high, VI-very high. First classification is for flow flow and second for high flow. Water-use problems: A blank indicates that no problem is known to exist. Listings under "Public supply" are based on the recommended limits from hazards under "irrigation supply" are based on U.S. Salinity Laboratory Staff (1954, p. 80). Boron: Range from samples taken. Remarks: Additional sediment data are shown on table 3 for each site except 16, 19.1, 54, 55. A (-) indicates no data available.

	Gaging			Discharge		Number of	Dissolved	Specific
Site No.	Station no.	Site name	Years of record	Range (ft ³ /s)	Average (ft ³ /s)	chemical analyses		conductance range (umho)
1	09415000	Virgin River at Littlefield, Arizona	52	38-35,200	234	5	1,120-2,620	1,620-3,500
2 3		Bear Dam Wash at mouth, Arizona Virgin River above Beaver Dam Wash, Arizona		3.7-5.9 61-318		5 5	479- 652 1 ,120-2, 720	
6	09413200	Virgin River near Bloomington, Utah	4	5.8-10,000	372	5	565- 2,550	870-3,450
7		Santa Clara River at mouth, Utah		4.5-9.7		5	1,150-1,800	1,600-2,180
8	09410100	Santa Clara River below Winsor Dam,	9	0-1,700	30.6	3	243-313	415-550
2		Utah Santa Clara River below Veyo Hot Springs	,	2.4-28		5	155-320	255-510
3		Utah Santa Clara River above Veyo Hot Springs	,	1.7-31		5	155-312	250-510
5		Utah Fort Pierœ Wash at mouth, Utah		2.1-36		5	1,110-2,690	1,510-4,020
		Virgin River above Fort Pierce Wash, Uta	h —	32-300		5	562-3,200	890-4,390
,	0 94 08150	Virgin River near Hurricane, Utah	14	23-18,700	237	5	524-2,670	860-3,120
З		Leeds Creek at mouth, Utah		1.2-3.4		5	901-1,200	1,180-1,540
9.1		Virgin River above Leeds Creek, Utah		69-500		5	511-2,680	815-3,140
		Virgin River below Ash Creek, Utah		55-400		5	531-2,280	905-3,7 00
2 7		Ash Creek at mouth, Utah La Verkin Creek at mouth, Utah		7.0-14 1.9-19		5 5	492 558 791-1,4 70	750-840 1,100-1,930
9		Virgin River below La Verkin Hot Springs	,	46-610		5	492-2,760	850-4,430
		Utah Virgin River above La Verkin Hot Springs. Utah	,	34-600		5	277-603	480-95 0
23	09406000	Virgin River at Virgin, Utah North Creek at mouth, Utah	65	22-22,800 .85-20	206	5 5	238-598 291-1,110	430-96 0 500-1,420
.1		Virgin River above North Creek, Utah		62-600		5	239-556	425-880
3		Virgin River below confluence of North Fork and East Fork, Utah		70-750	<u> </u>	5	17 9 -457	355-740
)		North Fork Virgin River above confluence		33-720		5	192-486	340-820
	09405500	North Fork Virgin River near Springdale, Utah	56	20-9,150	102	4	173-447	335-800
.2		North Fork Virgin River at mouth of Zion Narrows, Utah		44-700		3	164-333	325-580
	09405450	North Fork Virgin River above Zion Narrows, near Glendale, Utah	3	2.2-130	27.6	4	222-278	360-455
		Cascade Springs near Hatch, Utah		1.0-6.0		3	56-139	185-220
		Navajo Lake east of dike, near Hatch, Utah				3	53-77	97-135
		East Fork Virgin River above confluence		39-58		5	361-529	560-695
		with North Fork, Utah East Fork Virgin River at Mt. Carmel Instion Utab		3.8-22		5	421-788	670-1,160
	09404450	Junction, Utah East Fork Virgin River near Glendale,	15	6.3-640	20.9	5	231-331	500-580
7		Utah East Fork Virgin River at Highway 136 bridge, Utah		1.4-2.1		5	247-284	450-475

hydrlogic data at key sites

at $25^{\rm O}$ Celsius; mg/L, milligrams per liter; ug/L, micrograms per liter]

numbers indicate the downstream order of the stations.

magnesium; Na, sodium; HCO3, bicarbonate; Cl, chloride; SO4, sulfate.

U.S. Environmental Protection Agency (1976a, p. 5); Cl, chloride; N, nitrogen; NO3, nitrate; SO4, sulfate. Listings of salinity and sodium

Dominant	cation(s)	Dominar	nt anion(s)		Water-use	problems			
Low	High	Low	High	Public	Supply	Irrigation	Supply		
flow	flow	flow	flow	Dissolved solids (mg/L)	Specific ions (mg/L)	Sal inity hazard	Sodium hazard	Boron (ug/L)	Remarks
Ca, Na, Mg	Ca, Na, Mg	504,Cl	504, C1, HOO3	500	SO ⁴ , Cl 250	VH, H	M, L	370-1,100	Sediment data on table 1
Ca,Na,Mg Ca,Na,Mg	Ca,Na,Mg Ca,Na,Mg	ω_4, ω_3	HCO3, SO4	500		н, н	L, L	20-290	
-		so ₄ , ci	504, C1, HCO3	500	SO4,C1 250	VH, H	L, L	390-1,200	
Na,Ca	Ca, Na	SO4, Cl	SO4, HOO3	500	SO4,Cl 250	VH, H	M, L	140-1,100	
Ca, Mg	Ca,Mg	SO4, HCD3	SO4, HCO3	500	SO4 250	н, н	L, L	270-460	
Ca,Mg	Ca,Mg	H003, SO4	HCO3			м, м	L, L	40-70	Dry October 1981, February 1982
Ca,Mg	Ca,Mg	насу	HCO3			м, м	L, L	30-110	
Ca,Mg	Ca, Mg	HCD3	нсоз			м, м	L, L	30-70	
Ca,Na	Ca,Na,Mg	SO4,Cl	504,Cl	500	SO4,C1 250	VH, VH	M, L	190-960	
Va, Ca	Ca,Na	SO4,Cl	504,H003,Cl	500	SO4, C1 250	VH, H	M, L	130-1,200	
Ca,Na	Ca,Na	C1, SO4	HCO3, SO4, C1	500	504,C1 250	VH, H	M, L	3 50-950	Sediment data on table 1,2
Ca,Mg	Ca,Mg	SO4, HCO3	SO4, HCO3	500	SO4 250	Н, Н	L, L	230-270	
Na,Ca	Ca, Na	C1,HCO3	H003, SO4, CL	500	SO4, CL 250	VH, Н	M, L	130-920	
la,Ca	Ca, Na	SO4,Cl	HCO3, SO4, CL	500	SO4,Cl	VH, H	M, L	160-1,000	
Ca,Mg Ca,Mg	Ca,Mg Ca,Mg	SO4, HCO3	SO4, HCO3	500	250	н, н	L, L	60-80	
	-	SO4	SO4, HOO3	500	SO4 250	н, н	L, L	90-200	
la, Ca	Ca,Na	C1, SO4	HCO3, C1, SO4	500	SO4, Cl 250	VH, Н	н, L	150-1,300	
Ca,Mg,Na	Ca, Mg	H003,SO4	HCO3, SO4	500		н, н	L, L	30-80	
∴a,Mg,Na Ca,Mg	Ca,Mg Ca,Mg	SO_4 , HOO_3 SO_4 , HOO_3	HCO3, SO4 SO4, HCO3	500 500	SO4 250	Н, М Н, Н	L, L L, L	30-80 60-300	Sediment data on table 1,2
Ca,Na,Mg Ca,Mg,Na	Ca,Mg Ca,Mg	SO4,HCO3 HCO3,SO4	HCO3,SO4 HCO3	500		Н, М М, L	L, L L, L	10-70 20-60	
a,Na,Mg	Ca, Mg	C1,HCO3,SO4	нооз			н, м	L, L	20-50	
	Ca, Mg	HCO3, C1, SO4	нооз	*-	÷	н, м	L, L	10-40	
a,Mg,Na	Ca, Mg	HCO3, SO4	нас,			м, м	L, L	20-30	
a, Mg	Ca, Mg	н∞37204 н∞3	HCO3			M, L	L, L	20 20	Sediment data on table l
a,Mq	Ca, Mg	з нсоз	н003			L, L	L, L	0-10	Inaccessible February 198 Inaccessible February, May
a, Mg	Ca, Mg	HCO3	HCO3			L, L	L, L	0-10	1982 Inacœssible February, May
-	Ca, Mg	нсоз, SO4	H003, SO4			ы, ы м, м	L, L	40-7	1982
a, Mg	Ca, Mg	•			~				
-	-	SO ₄ , HCO ₃	HCO3, SO4	500	904 250	н, м	L, L	30-140	
	Mg, Ca	HCCC3	HCO3			м, м	L, L	20-30	
a,Mg	Ca,Mg	нсоз	HCO3			м, м	L, L	20-30	

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Table 8.--Chemical analyses and suspended-sediment loads of

(Abbreviations: ft³/s, cubic feet per second; °C, degrees Celsius; tons/d, tons per day; unito.

Other data available: F, fluvial sediment (table 3); K, key sites, additional hydrologic (table 7); P, pesticides (table 6); S, Statistical data

Site No	Site Name	Date of collection	Discharge (ft ³ /s)	Suspended sediment (tons/d)	Temperature (^O C)	рн	Specific conductance (umho)	Dissolved solids, sum of constituents (mg/L)	Dissolved silica (SiO ₂) (mg/L)	Dissolved calcium (Ca) (mg/L)	Dissolved magnesium (Mg) (mg/L)
1	Virgin River at Littlefield, Arizona	8-28-81 10-21-81 2-10-82 5-10-82 9- 3-82	71 136 205 322 80	627	26.0 18.0 10.5 19.5 22.0	7.0 7.8 8.2 7.9 7.7	3,500 2,990 2,580 1,620 3,320	2,620 2,150 1,870 1,120 2,300	20 16 20 13 20	400 320 240 163 340	120 96 76 44 110
2	Beaver Dam Wash at mouth, Arizona	8-28-81 10-21-81 2-10-82 5-10-82 9-3-82	3.7 5.9 5.0 4.0 5.2	 	21.5 17.5 17.5 20.0 21.5	7.4 7.9 7.5 7.8 7.6	980 770 805 750 840	652 529 479 523 538	38 39 37 39 36	95 87 60 81 80	28 25 23 24 24
2.1	Beaver Dam Wash below Bull Valley Wash, Utah	8-28-81 5-11-82	3.4 Dry		22.0	7.8	610	455	37	77	21
2,15	5 Bull Valley Wash at mouth, Utah	8-28-81 5-11-82	Dry Dry	Ξ							
2.2	Beaver Dam Wash at Motoqua, Utah	8-18-81 5-11-82	2.0 2.0	_	24.5 16.0	7.8 8.4	690 445	425 319	38 31	74 58	23 16
2.25	5 East Fork Beaver Dam Wash at mouth, Utah	8-28-81 5-11-82	Dry Dry	Ξ	=	_				=	
2.3	Beaver Dam Wash at end of Motoqua Road, Utah	8-18-81 5-11-82	2.5 2.5	_	24.0 16.0	8.0 8.5	375 400	325 290	44 37	54 53	14 13
2.4	Beaver Dam Wash below Beaver Dam State Park, Nevada	8-18-81 5-7-82	3.2 5.3	-	18.0 19.0	8.2 8.5	305 270	218 223	54 42	36 55	6.5 7.6
3	Virgin River above Beaver Dam Wash, Arizona	8-28-81 10-21-81 2-10-82 5-10-82 9-3-82	61 138 195 318 70		29.0 16.5 9.5 19.0 21.5	7.2 7.8 8.1 7.8 7.7	3,620 3,130 2,390 1,620 3,670	2.720 2,200 1,840 1,120 2.790	18 16 19 14 18	420 320 250 172 420	130 98 77 46 130
4	Virgin River below Virgin Gorge Springs, Arizona	8-28-81 9-3-82	30 37	_	27.5 21.5	7.4 7.5	3,690 3,650	2,660 2.610	17 18	440 340	130 120
5	Virgin River at Atkinville Wash, Arizona	8-28-81 5-5-82	37 450	115	32.0 12.5	7.3 8.0	4,200 870	2,990 577	25 9.6	350 98	110 20
6	Virgin River near Bloomington, Utah	8-28-81 10-15-81 2-10-82 5- 5-82 9- 2-82	35 139 206 610 45	123 	29.0 15.5 7.0 15.0 31.0	8.0 8.0 8.2 7.9 7.9	4,180 2,630 1,990 870 3,540	2,830 1,780 1,340 565 2,550	24 16 18 11 24	360 230 160 97 320	100 66 53 22 100
7	Santa Clara River at mouth, Utah	8-18-81 10-15-81 2-11-82 5-5-82 8-23-82	4.5 5.1 4.5 9.7 5.2	1.8 	26.0 16.0 8.5 18.5 28.0	7.8 7.9 8.3 7.9 8.0	1,960 2,180 2,150 1,600 1,890	1,430 1,600 1,680 1,150 1,420	36 32 36 29 37	256 300 310 215 260	70 94 83 55 68
8	Santa Clara River below Winsor Dam near Santa Clara, Utah	8-18-81 10-15-81 2-11-82 5-4-82 8-23-82	18 Dry Dry 20 11	2.0 	24.5 15.5 24.5	8.2 8.3 8.6	415 445 550	243 282 313	26 26 26	43 52 50	16 17 16
8.1	Santa Clara River below Gunlock Reservoir, Utah	8-18-81 5-4-82	28 32		19.5 11.0	7.8 8.1	435 425	259 263	26 27	46 50	16 17
9	Santa Clara River at Gunlock. Utah	8-19-81 5-4-82	6.2 34	-	26.0 17.0	8.0 8.2	46 0 320	273 188	37 25	47 37	17 12
10	Moody Wash at mouth, Utah	8-20-81 5-4-82	1.2 4.3		23.0 18.5	7.6 7.9	465 440	2 88 277	48 44	50 52	16 15
11	Magotsu Creek at mouth, Utah	8-17-81 5- 4-62	.17 2.6		25.5 17.5	7.9 8.1	780 700	469 413	4 2 35	72 72	32 26
12	Santa Clara River below Veyo Hot Springs, Utah	8-18-81 10-15-81 2-12-82 5- 6-82 8-24-82	2.6 2.4 3.5 32 2.7	14 	25.0 20.3 16.0 12.5 24.5	7.7 8.1 8.4 8.1 8.0	510 475 495 250 470	286 315 320 155 294	35 34 33 22 31	53 54 56 30 51	20 19 20 9.6 17

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water samples collected in the Virgin River basin, 1981-82,---Continued

per centimeter at 25°C; mg/L, milligrams per liter; ug/L, micrograms per liter}

(table 9); T, trace metals (table 1)

Dissolved sodium (Na) (mg/L)	Dissolved potassium (K) (mg/L)	Alkalinity (total as CaCo ₃) (mg/L)	Dissolved sulfate (SO4) (mg/L)	Dissolved chloride (Cl) (mg/L)	Dissolved fluoride (F) (mg/L)	Dissolved nitrate (NO ₃)+ nitrate (NO ₂) as (MG/L)	Phosphorus, ortho- phosphate N as P (mg/L)	Total hardness (as CaCO3) (mg/L)	Non- carbonate hardness (as CaCo ₃) (mg/L)	Sodium- absorption ratio	Dissolved boron (B) (uq/L)	Dissolved oxygen (mg/L)	Other data available
270 270 250 130 290	27 23 20 12 27	240 240 270 220 251	1,200 880 760 440 990	430 400 340 180 370	0.9 .8 .8 .4 .8	1.1 .86 .85 .44 .90	0.02 <.01 .05 .03	1,500 1,200 910 590 1,300	1,300 950 640 370 1,100	3.0 3.4 3.6 2.5 3.5	1,100 850 650 370 1,100	7.1 8.5 90 9.2 7.6	F. K. S
72 51 45 55 59	5.3 6.6 5.4 4.7 5.0	200 230 230 220 210	240 140 130 150 170	44 32 12 31 32	.5 .6 .7 .7 .6	1.9 2.2 1.7 1.2 1.2	.03 <.01 .01 .02 .01	350 320 290 300 300	150 90 65 81 89	1.8 1.3 1.2 1.5 1.6	290 20 60 130 160	7.4 6.0 8.0 5.9 5.7	к. т
36	4.3	200	140	18	.7	.13	<u>.02</u>	280	79	1.0	60	8-2	
					-								
29 22	4.0 3.3	190 170	110 68	31 18	.5 .6	.19 <.10	.03 .01	2 80 210	90 41	.8 .7	50 30	6.3 10.4	
26 21	4.7 3.6	130 150	81 59	22 13	.6 .6	.12 <.10	.02 .01	190 190	63 36	.9 .7	30 20	82 9.0	
19 19	5.8 5.0	130 130	<5.0 6.0	15 10	.5 .5	.17 <.10	.02 .02	120 170	0 39	.8 .7	30 30	80 9.7	
290 280 250 130 300	30 24 20 12 29	240 200 290 220 269	1,200 920 720 430 1,300	480 420 320 180 430	.8 .7 .4 .9	1.2 .87 .88 .46 .85	.02 <.01 .05 .02 <.01	1,600 1,200 940 620 1,600	1,300 1,000 650 400 1,300	3.2 3.5 3.5 2.4 3.3	1,200 880 660 390 1,100	7.2 8.2 10.6 9.1 7.9	ĸ
280 300	28 30	160 287	1,200 1,200	470 430	.9 .8	.33 .42	<.02 .01	1,600 1,300	1,500 1,100	3.0 3.6	1,000 1,000	6.9 7.3	т
450 60	28 61	240 150	1,200 220	670 71	.6 .2	1.4 .31	.09 .03	1,300 330	1,100 180	5.4 1.5	1,100 140	7.2 9.4	
460 280 200 55 390	31 20 16 6.1 26	250 240 250 160 260	1,000 620 480 210 980	700 400 260 66 550	.7 .5 .3 .7	1.5 .98 .87 .36 .90	.07 .02 .13 .04 .21	1,300 850 620 330 1,200	1,100 610 370 170 950	5.5 4.2 3.8 1.4 4.9	1,100 600 450 140 950	7.0 8.5 10.9 9.9 6.3	К. Т
110 130 110 72 99	7.5 11 9.4 6.8 8.2	240 260 320 270 242	750 990 870 560 740	55 81 62 51 62	.3 .5 .5 .4 .4	.30 1.2 1.1 .38 .39	.01 .01 .01 .03 .03	930 1,100 1,100 760 930	690 880 800 490 690	1.7 1.7 1.4 1.2 1.5	370 460 370 270 370	7.9 11.3 10 2 9.8 9.5	К, Р. Т
19	2.8	160	23	16	.1	.12	.02	170	13	.7	60	6.7 	К
19 19	2.9 3.0	1 80 177	 35 78	22 25	 .3 .2	.10 <.10	.01 .02	200 190	20 32	 .6 .6	 40 70	8.4 7.4	
18 18	2.8 2.8	160 170	24 24	30 22	.2 .2	.03 <.10	.09 .01	180 190	21 25	.6 .6	50 60	7.6 9.3	
20 12	3.5 2.0	170 130	21 7.0	23 15	.2 .2	.28 <.10	.02	190 140	18 12	.7 .5	60 50	7.8 8.9	
23 21	2.4 2.0	160 190	1.0 6.0	50 22	.2 .3	.15 <.10	.02 .02	190 190	31 2	.8 .7	50 40	7.0 7.3	
45 36	5.4 4.5	250 250	53 45	69 44	.3 .3	<.10 <.10	.02 .02	310 290	62 37	1.2 1.0	110 8.9	7.2 0 7.5	
22 21 21 9.7 1.9	4.5 5.1 5.3 2.0 4.4	150 200 190 110 196	31 29 33 6.0 26	24 28 33 9.9 24	.3 .4 .4 .1 .3	1.3 .89 .88 <.10 .80	.03 .02 .01 .03 .02	210 210 220 120 200	65 13 32 5.0 1.0	.7 .7 .4 .6	90 60 110 30 80	6.6 6.3 7.8 8.9 6.4	т, К

Table 8.---Chemical analyses and suspended-sediment loads of

Site No	Site Name	Date of collection	Discharge (ft ³ /s)	Suspended sediment (tons/d)	Temperature (^O C)	pH	Specific conductance (umho)	Dissolved solids. sum of constituents (mg/L)	Dissolved silica (SiO ₂) (mg/L)	Dissolved calcium (Ca) (mg/L)	Dissolved magnesium (Mg) (mg/L)
12.1	Veyo Hot Springs, Utah	8-18-81 5-6-82	.67 .75		31.0 16.0	7.1 7.9	650 630	40 4 387	38 35	58 56	27 27
13	Santa Clara River above Veyo Hot Springs, Utah	8-18-81 10-15-81 2-12-82 5-6-82 8-24-82	1.9 1.7 2.9 31 2.1	1,322	24.5 19.6 13.0 12.5 24.5	7.9 8.1 8.6 8.1 8.0	490 510 490 250 450	297 293 312 155 290	35 34 32 22 31	53 54 57 30 54	18 18 19 9.6 17
13.1	Santa Clara River below Baker Dam Reservoir, Utah	8-17-81 5-4-82	3.7 11		21.0 10.0	7.6 8.0	360 265	217 162	36 22	38 32	14 9.6
13.2	Santa Clara River above Baker Dam Reservoir, Utah	8-17-81 5- 4-82	5.4 64		17.5 9.0	7.6 7.8	295 82	180 64	37 16	33 9-2	10 3.1
13.3	Spring above Baker Dam, Utah	8-17-81 5-4-82	2.8 4.3		16.5 16.5	7.2 7.7	305 290	187 197	37 36	34 35	11 11
14	Santa Clara River near Pine Valley, Utah	8-17-81 5- 4-82	3.2 58	_	15.5 4.0	8.1 7.6	96 44	67 41	20 13	9.3 4.7	4. 2 1.8
14.1	Santa Clara River below Right Fork, Utah	8-18-81 5-4-82	2.8 58	_	11.0 3.5	7.1 6.8	95 44	75 40	23 12	9.1 5.4	4.2
14.2	Santa Clara River above campground, Utah	8-19-81 5-4-82	.86 36		11.5 2.5	7.1 6.9	120 37	90 36	25 11	12 4 .0	5.1 1.4
14.3	Santa Clara-Pinto Diversion near Pinto, Utah	8-18-81 5-4-82	Dry 72		3.0	 7.7	 65	 58		7.8	1.9
15	Fort Pierce Wash at mouth, Utah	8-28-81 10-16-81 2-10-82 5- 5-82 9- 2-82	2.1 36 17 24 6.0	43 	30.0 10.8 6.0 21.0 29.0	7.7 8.0 8.4 7.9 7.9	4,020 2,300 1,930 1,510 3,210	2,690 1,460 1,260 1,110 2,250	21 15 16 9-4 20	410 180 150 227 300	85 53 48 36 67
15.1	Fort Pierce Wash at bottom of section 16, Utah	8-28-81 5- 5-82	Dry 40		28.0	 7.6	2,690	2,650	 6.5	 580	 85
15.2	Short Creek at Temple Trail, Arizona	8-28-81 5-5-82	Dry 1.3	 8.4	21.0	7.4	2,440	2,350	 4.9		 58
15.25	Hurricane Wash above confluence with Short Creek, Arizona	8-28-81 5-5-82	Dry Dry	_							
15.3	Short Creek at Highway U-59 Arizona	8-28-81 5-5-82	Dry Dry	_			-				
16	Virgin River above Fort Pierce Wash, Utah	8-28-81 10-16-81 2-11-82 5- 5-82 9- 1-82	29 97 162 575 34		32.0 14.0 7.0 15.0 24.0	7.7 7.9 7.9 8.0 7.8	4.390 2,900 1,780 890 3,400	3.200 1,870 1.200 562 2.510	26 16 16 9.4 20	380 240 140 93 310	120 72 50 21 99
16.1	Mill Creek below Washingon, Utah	8-20-81 5-6-82	1.9 2.9	=	27.5 14.0	7.9 8.1	1,400 1,480	852 936	23 24	91 97	26 31
16.2	Washington Canal at inlet, Utah	8-24-81 5-6-82	87 175	2,555	23.0 15.0	7.6 8.2	3,450 820	2,850 501	16 9.5	560 82	64 20
17	Virgin River near Hurricane, Utah	8-31-81 10-20-81 2-11-82 5-6-82 8-24-82	93 124 220 550 155	449 2,510 34,108	20.0 12.5 9.0 16.0 28.0	7.9 7.9 8.1 8.1 7.9	3,120 1,930 1,560 860 2,710	2,000 1,310 1,010 524 2,670	20 12 15 9.9 12	190 150 125 86 620	59 49 42 21 45
18	Leeds Creek at mouth, Utah	8-19-61 10-20-61 2-11-62 5-7-62 8-23-62	1.2 1.7 3.4 1.9 1.5	81 21	25-5 10.5 12-0 13.0 29.5	7.9 8.1 8.2 8.0 8.0	1,540 1,450 1,180 1,510 1,500	1,100 1,040 901 1,110 1,200	30 28 26 27 27	170 160 123 174 180	85 81 67 84 90
19	Leeds Creek near Leeds, Utah	8-24-81	3.0		16.3	8.3	255	175	30	37	12
19.1	Virgin River above Leeds Creek, Utah	8-31-81 10-20-81 2-11-82 5- 7-82 8-23-82	69 125 215 550 153	304 1,886	20.0 12.5 9.0 11.0 27.5	7.9 8.0 8.0 8.0 7.9	3,140 2,110 1,570 825 2,700	1,820 1,360 1,220 511 2,680	20 14 14 9.8 11	200 170 117 8.6 610	55 50 39 21 43
20.1	Gould Wash at mouth, Utah	8-28-81 5-7-82	1.4 2.4		28.0 15.5	8.5 8,2	900 575	567 369	14 9-0	92 82	2 9 17

water samples collected in the Virgin River basin, 1981-82 .-- Continued

Dissolved sodium (Na) (mg/L)	Dissolved potassium (K) (mg/L)	Alkalinity (total as CaCog) (mg/L)	Dissolved sulfate (SO4) (mg/L)	Dissolved chloride (Cl) (mg/L)	Dissolved fluoride (F) (mg/L)	Dissolved nitrate (NO ₃)+ nitrate (NO ₂) as (mg/L)	Phosphorus, ortho- phosphate N as P (mg/L)	Total hardness (as CaCO ₃) (mg/L)	Non- carbonate hardness (as CaCo ₃) (mg/L)	Sodium- absorption ratio	boron (B) (ug/L)	Dissolved oxygen (mg/L)	Other data available
33 29	3.7 3.7	210 200	79 77	27 30	.3 .4	2.7 1.9	.03 .02	260 250	46 51	1.0 .9	150 120	46 7.0	Т
19 19 18 9.1 18	4.6 5.2 5.6 2.0 4.8	200 190 180 110 196	16 20 27 6.0 21	27 25 41 9.8 23	.3 .4 .1 .3	.94 .75 .74 <.10 .68	.03 .02 .02 .02 .02	210 210 220 110 200	6 19 41 4.0 9-0	.6 .6 .4 .6	70 50 60 30 70	6.5 7.0 8.0 9.0 6.4	Т, К
13 8.7	2.1 1.7	140 120	5.0 6.0	23 9.5	.1 .1	.27 <.10	.08 .03	150 120	13 0	.5 .4	30 10	5.9 8.3	
10 3.2	1.9 .8	130 39	<1.0 6.0	8.8 1.8	.1 <.1	.35 <.10	.05 .04	120 36	0 0	.4 .3	20 10	7.7 9.4	
· 11 11	2.0 2.3	130 140	1.0 6.0	11 9.9	.1 .1	.49 .32	.04 .05	130 130	0 0	.5 .4	20 20	7.8 7.6	
4.3 1.9	.5 .4	40 21	2.0	2.1	.1 <.1	.11 <.10	.03 .03	41 19	1.0 0	.3 .2	10 10	8.0 10.4	
4.1 1.8	.4 .4	4 6 20	5.0 6.0		.2 <.1	0.0 <.10	.21	40 20	0 0	•3 •2	10 10	8.4 10-2	
5.0 17	.4 .4	62 18	2.0 6.0	2.6	.0 <.01	.14 .10	.06 .02	51 16	0 0	.3 .2	20 10	8 .4 10-7	
2.9	 1.0	31	 6.0	1.9	<.1	.10	.04	27	0	.3	10	10-4	
3 80 240 190 6 8 350	25 17 15 11 24	200 200 200 130 168	1,000 470 450 600 870	640 360 270 73 510	.4 .4 .2 .5	1.2 .68 .64 .69 .96	<.02 <.01 .01 .03 .02	1,400 670 570 720 1,000	1,200 470 370 590 860	4.5 4.0 3.7 1.2 4.8	960 520 400 190 700	6.6 10.0 11.6 8.2 6.9	Т. К Т
73	20	 59	1,800	 45	.2	2.0	.03	 1,800	 1,700	.8	 2 90	 6.0	
 46	15	 51	1,600	31	.2	1.7	.02	1,600	1.600	 .5	180	7.8	
470 290 180 59 390	33 21 15 6.1 28	260 170 210 160 286	1,400 700 410 200 950	610 420 260 76 540	.8 .5 .4 .2 .7	1.6 .73 .67 .29 .88		1,400 900 560 320 1,200	1,200 730 350 160 900	5.4 4.2 3.6 1.5 4.9	1,200 670 390 130 910	6.4 8.9 10.4 10.8 7.1	Р, К
150 160	16 16	1 9 0 210	270 300	160 180	.7 .9	.14 .21	.33 .03	330 370	140 160	3.8 3.9	330 360	5.6 9.0	т
250 55	21 5.4	160 180	1,500 140	340 79	.3 .2	.68 .29	<.01 .02	1,700 290	1,500 110	2.7	660 130	6.1 11.0	
400 230 150 65 150	27 17 14 5.6 15	230 190 180 180 178	570 400 330 140 1,500	590 330 220 87 220	.5 .4 .2 .3	.85 .60 .61 .30 .48	.04 <.01 .02 .02 <.01	720 580 490 300 1,700	490 390 310 120 1,600	6.5 4.5 3.2 1.7 1.6	950 490 320 140 350	7.9 9.1 10.5 10.3 6.6	F. K
64 60 58 65 65	4.8 4.9 8.2 5.4 5.0	250 230 280 280 214	550 530 420 550 670	41 35 27 31 35	.3 .4 .5 .4	.90 .74 .57 .52 .52	<.02 .01 <.01 .01 <.01	770 730 580 780 820	520 500 300 500 610	1.1 1.1 1.2 1.1 1.1	260 240 230 280 270	7.1 9.6 10.0 9.2 8.0	Т, К
6.4	.9	140	<5.0		.1	.12	.05	140	2	.3	40	8.2	
380 230 160 60 160	28 17 14 4.9 16	350 210 230 190 187	260 400 310 130 1,500	660 350 220 83 220	.5 .4 .2 .3	.75 -58 -28 -43	<.02 <.01 .01 .02 <.01	730 630 460 300 1,700	380 420 230 110 1,500	6.1 4.0 3.2 1.6 1.7	920 500 330 130 370	8.0 9.3 10 8 10.4 6.7	Т, Р, К
51 17	8.9 2.9	190 170	200 120	56 17	.2 .1	.44 .30	.15 .05	350 270	160 100	1.3 .5	80 30	6.1 8.6	т

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Table 8.—Chemical analyses and suspended-sediment loads of

Site No	Site Name	Date of collection	Discharge (ft ³ /s)	Suspended sediment (tons/d)	Temperature (^O C)	рн	Specific conductance (umho)	Dissolved solids. sum of constituents (mg/L)	Dissolved silica (SiO ₂) (mg/L)	Dissolved calcium (Ca) (mg/L)	Dissolved magnesium (Mg) (mg/L)
21	Virgin River below Ash Creek, Utah	8-25-81 10-20-81 2-10-82 5-7-82 9-2-82	55 128 239 510 89	254 	29.0 13.5 5.5 12.5 23.0	7.8 7.8 7.7 7.6 7.9	3,700 2,290 1,900 905 2,840	2.280 1,410 1,220 531 1,770	20 14 14 8.6 16	240 160 135 84 200	60 52 42 19 53
22	Ash Creek at mouth, Utah	8-25-81 10-20-81 2-10-82 5- 7-82 9- 2-82	7.0 13 14 11 11	1.4 	19.5 15.0 10.0 17.5 20.0	7.8 8.2 8.4 8.1 8.4	840 750 750 705 790	558 541 499 492 533	40 41 39 38 39	97 86 83 82 88	39 37 34 33 38
23	Ash Creek below West Field, Ditch at Toquerville, Utah	8-25-81 5- 7-82 9- 2-82	9.3 7.5 4.0		16.5 17.5 16.5	7.6 8.2 7.8	700 625 700	450 431 439	42 3 8 3 8	79 72 74	31 28 29
23.1	Wet Sandy at Anderson Ranch, Utah	8-25-81 5-11-82	Dry Dry								
24	South Ash Creek at mouth, Utah	8-25-81 5-11-82	Dry 12		10.0	8.7	175	105	21	21	 5.1
25	South Ash Creek below Mill Creek near Pintura, Utah	8-17-81 5-11-82	1.6 12	-	15.5 5.0	8.6 8.1	185 175	56 106	25 22	26 21	6.5 5.3
25.1	Leap Creek near Pintura, Utah	8-17-81 5-11-82	Dry 2.0	_	9.0	8.5	 365	 240	27	 45	15
25.2	North Ash Creek below Ash Creek Reservoir, Utah	8-18-81 5-11-82	Dry Dry	Ξ	<u> </u>		=				_
26	North Ash Creek above Ash Creek Reservoir, Utah	8-18-81 5-11-82 9- 2-82	1.5 19 1.3	=	24.5 10-5 12.5	8.0 8.5 8.3	505 410 750	311 261 4 <i>8</i> 2	48 32 38	58 46 92	20 16 35
26.1	Kanarra Creek at Kanarraville, Utah	8-17-81 5-11-82	2.6 7.2	_	12-0 7.0	8.5 8.0	415 475	284 278	24 14	56 66	17 16
27	La Verkin Creek at mouth, Utah	8-25-81 10-20-81 2-10-82 5- 7-82 9- 2-82	2.2 5.2 7.5 19 1.9	44 	25.0 14.5 2.0 20.0 26.0	8.0 8.0 8.1 8.2 7.9	1,520 1,250 1,100 930 1,930	1,090 932 791 654 1,470	21 16 11 11 19	228 177 152 130 300	56 50 43 38 63
29	Virgin River below La Verkin Hot Springs, Utah	8-25-81 10-20-81 2-10-82 5-5-82 9-2-82	46 70 136 510 65	143 2.750	30.0 15-0 8.5 13-5 24.0	6.8 6.8 6.5 8.3 6.8	4,430 2,780 2,780 850 3,320	2.760 1,740 1.760 492 2,060	16 11 13 9.1 13	280 200 184 78 210	43 51 48 17 53
30	La Verkin Hot Springs, Utah	8- 25-81 5- 5-82	11 11		41.5 41.5	6.4 6.1	13,000 12,600	9-660 9,840	28 24	820 880	160 160
31	Virgin River above La Verkin Springs, Utah	8-25-81 10-20-81 2-10-82 5- 5-82 9- 2-82	34 77 151 500 53	79 3,289	26-5 10.5 3.5 12.5 19.0	8.2 8.3 8.1 8.4 8.5	910 950 820 4 80 81 0	560 603 498 277 504	12 7.0 10 8.1 11	92 110 77 60 80	30 34 30 14 30
32	Virgin River at Virgin, Utah	8-24-81 10-17-81 2-10-82 5-5-82 9-1-82	80 123 156 620 82	834 	29.5 15-5 5.5 10-5 17.0	8.2 8.3 8.2 8.3 8.1	960 850 850 430 840	598 564 526 238 489	12 7.2 10 7.6 12	106 94 81 55 79	28 35 33 13 31
33	North Creek at mouth, Utah	8-24-81 10-17-81 2-10-82 5- 5-82 8-23-82	.85 3.6 6.7 20 1.5	.06 	30.5 12.5 5-5 16.0 26.0	8.1 8.2 8.3 8.4 8.2	1,350 1,420 1,160 500 1,380	958 1,020 806 291 1,110	20 18 15 10 20	170 200 140 57 210	47 47 38 15 56
33,1	Virgin River above North Creek, Utah	8-24-81 10-17-81 2-10-82 5- 5-82 9- 1-82	62 110 133 600 80	473	28.5 15-5 6.5 11.5 17.0	8.2 8.3 8.5 8.4 8.5	880 81.0 800 425 790	556 523 469 239 466	12 12 9.9 8-0 11	89 82 71 55 69	27 32 31 14 29
86	Coalpits Wash at mouth, Utah	8-24-81 5- 5-82	Dry Dry								

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water samples collected in the Virgin River basin, 1981-82,---Continued

Dissolved sodium (Na) (mg/L)	Dissolved potassium (K) (mg/L)	Alkalinity (total as CaCo ₂) (mg/L)	Dissolved sulfate (SO ₄) (mg/L)	Dissolved chloride (Cl) (mg/L)	Dissolved fluoride (F) (mg/L)	Dissolved nitrate (NO ₃) + nitrate (NO ₂) as N (mg/L)	Phosphorus, ortho- phosphate as P (mg/L)	Total hardness (as CaO3) (mg/L)	Non- Carbonate hardness (as CaCo ₃) (mg/L)	Sodium⊢ absorption ratio	Dissolved boron (B) (ug/L)	Dissolved oxygen (mg/L)	Other data available
450 260 210 68 350	36 20 17 5.7 26	280 240 260 190 277	630 390 350 130 460	670 370 290 100 490	0.5 .5 .4 .2 .5	0.74 .55 .53 .27 .56	<0.02 <.01 <.01 .02 .01	850 610 510 290 720	570 370 250 98 440	6.7 4.6 4.3 1.8 5.7	1.000 540 440 160 820	6.3 9.6 11.4 9.5 8.1	Т, К
26 26 21 21 24	3.1 3.4 3.3 2.5 3.0	200 190 180 190 196	220 210 190 180 200	11 20 16 18 19	.2 .2 .2 .2 .2	.31 .77 .87 .69 .85	.01 <.01 .01 .02 .02	400 370 350 340 380	200 180 170 150 180	.6 .5 .5 .6	70 60 60 60 80	7.4 9.4 9.9 7.8 7.7	К
21 18 19	2.2 2.4 2.9	110 140 141	170 160 170	35 25 18	.1 .2 .2	.90 .62 .75	.04 .01 .02	330 300 300	220 160 160	.6 .5 .5	50 50 60	7.9 8.3 8 2	g
			-1-										
3.5	.4	 79	 <5.0	1.5	<.1	<.10	.03	73	0	.2	10	96	
4.4 3.8	.6 .3	92 78	1.0 <5.0	3.6 1.6	.1 <.1	.10 <.10	.00 .04	92 7 4	0 0	.2 .2	10 10	8.2 10 5	
	 .8	170	 5.0	 4.9	.1	<.10	.02	 170	 4 . 0	.3	30	9 2	
16 12 23	2.4 2.3 2.6	190 130 161	33 66 170	18 8.1 22	.1 .1 .2	.22 .17 .54	.02 .03 <.01	230 180 370	37 51 210	.5 .4 .6	30 30 60	6.7 96 9.2	
9.1 7.2	1.7 1.3	140 170	83 64	7.0 5.9	.1 .2	.38 .26	.03 .02	210 230	70 61	.3 .2	20 20	8 2 8.4	
42 40 33 22 47	4.8 4.4 3.7 3.9 5.9	150 150 180 160 132	630 530 420 340 920	17 23 18 12 32	.2 .2 .2 .2 .3	.18 .32 .44 .18 .57	.00 <.01 <.01 .01 <.01	800 650 560 480 1,000	650 500 380 320 880	.7 .7 .7 .5 .7	170 140 110 90 200	6.8 10.4 12.3 7.8 7.3	Т, К
580 330 350 63 440	43 26 29 8.7 32	210 300 310 170 329	690 460 430 120 480	960 4 80 520 95 630	.7 .5 .6 .2 .6	.48 .40 .48 .23 .32	.02 <.01 <.01 .02 <.01	960 710 660 260 740	750 410 350 95 410	8.2 5.4 6.3 1.8 7.0	1,300 680 790 150 1,000	5.2 9.1 9.4 8.5 6.4	к
2,300 2,300	150 150	2.0 1,020	2,100 2,100	4.100 3,600	2.8 2.9	.14 <.10	.06 .05	2.700 2.900	1,800	19 19	870 5,200	.5 1.1	T
51 46 51 18 50	4.6 3.9 4.4 2.4 4.2	160 160 170 140 159	230 250 180 72 180	43 54 41 17 51	.2 .2 .1 .2	.21 .48 .56 .23 .40	.05 <.01 <.01 .02 .02	350 410 320 210 320	190 250 150 67 160	1.3 1.1 1.4 .6 1.3	80 60 70 30 80	6.8 9.7 12 1 9.1 8.2	
53 51 50 10 48	5.0 4.5 4.5 2.0 4.0	150 170 180 140 150	240 210 180 55 170	61 58 57 10 53	.2 .2 .3 .1 .2	.62 .49 .47 .26 .41	.03 <.01 <.01 .03 <.01	380 380 340 190 320	230 210 160 51 170	1.3 1.2 1.3 .3 1.3	80 60 30 80	6.3 8.5 11.4 10 1 8.2	F. Т. К
54 53 15 57	8.0 5.8 5.3 2.1 9.0	150 160 170 99 132	530 560 430 120 640	37 37 18 12 40	.1 .2 .1 .2	.23 .60 .82 .13 .10	.03 <.01 <.01 .02 <.01	620 690 510 200 760	470 530 340 110 620	1.0 .9 1.1 .5 1.0	270 230 170 60 300	9.0 9.6 11.6 9.0 8.8	Т, К
52 49 48 10 48	5.1 4.5 4.3 1.9 4.0	170 180 170 150 178	230 180 150 49 150	26 53 50 10 46	.2 .3 .1 .2	.43 .48 .42 .23 .39	<.01 <.01 <.01 .02 .02	330 160 310 200 290	160 32 140 45 110	1.3 1.3 1.3 .3 1.3	60 50 50 10 70	6.2 8.6 11.0 11.0 8.2	ĸ

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Table 8.---Chemical analyses and suspended-sediment loads of

Site N⊍	Site Name	Date of collection	Discharge (ft ³ /s)	Suspended sediment (tons/d)	Temperature (^O C)	рН	Specific conductance (umho)	Dissolved solids, sum of constituents (mg/L)	Dissolved silica (SiO ₂) (mg/L)	Dissolved calcium (Ca) (mg/L)	Dissolved magnesium (Mg) (mg/L)
37	Horse Valley Wash at mouth, Utah	8-24-81 5- 4-82	Dry Dry	_							
38	Virgin River below confluence of North Fork and East Fork, Utah	8-24-81 10-20-81 2-11-82 5- 4-82 9- 1-82	70 93 126 750 80	83 1,869	26.2 9.0 4.0 9.5 17.0	8.3 8.3 8.2 8.4 8.5	740 650 670 355 740	418 457 429 179 415	11 11 95 6.7 11	60 62 58 47 60	27 30 28 9.6 27
39	North Fork Virgin River above confluence with East Fork, Utah	8-24-81 10-20-81 2-12-82 5- 4-82 8-23-82	33 42 50 720 65	39 1,928 7,125	24.5 8.5 2.5 8.5 21.5	8.0 8.3 8.2 8.4 8.4	820 720 780 340 520	461 470 486 192 352	11 11 9.5 7.1 8.5	59 63 61 47 58	26 26 9.6 17
39.1	Oak Creek below Park Headquarters, Utah	8-24-81 5-4-82	Dry Dry	_		_	_				
40	North Fork Virgin River near Springdale, Utah	8-24-81 2-11-82 5- 4-82 8-23-82	43 49 720 65	1,608 2,568	18.5 4.5 7.0 19.0	8.2 8.2 8.3 8.4	790 800 335 495	437 447 173 341	11 9.3 6.9 11	56 61 46 60	25 24 9-4 17
41	Pine Creek at mouth, Utah	8-24-81 5- 4-82	Dry Dry								
41.1	Clear Creek below Co-op Creek, near East Zion Entrance Station, Utah	8-24-81 5- 4-82	Dry Dry								
42	Birch Creek at mouth, Utah	8-24-81 5- 4-82	Dry Dry								
43.1	Weeping Rock at trail, Utah	8-19-81 5- 4-82	.70 .40	-	19-0 16.0	8.0 7.8	1,250 1,180	733 709	12 11	62 63	29 30
43 . 2	North Fork Virgin River at mouth of Zion Narrows, Utah	8-19-81 10-17-81 2-11-82 5-4-82	44 54 45 700	12	20-5 8.6 6.0 6.0	8.2 8.6 8.3 8.5	580 540 550 325	333 336 340 164	11 11 9.9 6.7	54 59 60 46	24 24 24 9.2
44.1	Ordervile Gulch at road crossing, Utah	8-20-81 5- 3-82	.70 .50		16.0 16.5	8.3 8.5	6 80 850	419 522	8-9 8.0	65 78	40 47
45	Kolob Creek below reservoir, Utah	8-24-81	Dry		-						
46	Kolob Creek above reservoir, Utah	8-17-81	.70	-	17.5	8.4	4 6 0	281	6.8	82	15
50	Deep Creek at road crossing 1 mile east of county line, Utal	82081 h	2.0		16.5	8,3	415	236	8.3	75	16
51	North Fork Virgin River above Zion Narrows near Glendale, Utah	8-20-81 10-8-81 5-3-82 8-23-82	7.5 7.3 34 13		16.0 14.5 14.5 22.0	8.4 8.5 8.0 8.4	420 435 455 360	250 278 246 222	12 12 85 12	54 55 53 45	24 26 23 22
52	North Fork Virgin River below Bullock Canyon near Glendale, Utah	8-20-82 5-3-82	9.3 42	-	14.5 12.0	8.2 8.1	385 435	237 222	12 7.9	54 54	22 21
54	Cascade Springs near Hatch, Utah	8-20-81 10-8-81 8-23-82	2.5 1.0 6.0	_	12.0 8.0 13.5	8.8 7.7 7.8	185 220 211	56 139 118	4 0 5.6 3.7	26 36 31	7.9 10 7.4
55	Navajo Lake east of dike near Hatch, Utah	8-20-81 10-12-81 8-23-82	=	=	18.0 3.0 20.0	9.0 9.3 8.6	105 97 135	53 63 77	.99 1.9 2.4	11 12 20	6.5 6.6 6.9
56	East Fork Virgin River above confluence with North Fork, Utah	8-24-81 10-20-81 2-12-82 5- 4-82 8-23-82	39 58 47 50 53	118 46	24.5 8.5 2.5 17.5 22.0	8.0 8.5 8.3 8.2 8.3	620 570 560 660 695	384 404 361 378 529	12 12 9.6 11 11	60 61 55 54 97	28 34 30 32 28
57	Meadow Creek at Highway U-15, Utah	8-24-81 5- 4-82	Dry Dry	Ξ		-	=				
50	East Fork Virgin River at Mount Caramel Junction, Utah	8-20-81 10-12-81 2-11-82 5-3-82 8-31-82	3.8 21 22 14 9.6	2.6 12 	23.5 13.0 4.5 18-5 19.0	8.0 8.3 8.2 8.4 8.1	1,160 920 670 870 1,110	7 68 635 421 539 705	11 11 9.0 92 11	118 112 76 82 110	71 55 46 55 64

water	samples	collected	in t	the Virgi	n River	basin,	1981-82Continued
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Dissolved sodium (Na) (mg/L)	Dissolved potassium (K) (mg/L)	Alkalinity (total as CaCo3) (mg/L)	Dissolved sulfate (SO4) (mg/L)	Dissolved chloride (Cl) (mg/L)	Dissolved fluoride (F) (mg/L)	Dissolved nitrate (NO3)+ nitrate (NO2) as I (mg/L)	Phosphorus, ortho- phosphate N as P (mg/L)	Total hardness (as CaOO3) (mg/L)	Non- carbonate hardness (as CaCo3) (mg/L)	Sodium- absorption ratio	Dissolved boron (B) (ug/L)	Dissolved oxygen (mg/L)	Other data available
48 52 48 9.2 47	4.0 4.6 3.3 1.6 3.6	160 180 190 130 168	130 130 100 13 110	41 57 66 13 54	0.2 .2 .3 .1 .2	0.21 .42 .41 .21 .33	0.02 <.01 <.01 .02 <.01	260 280 260 160 260	100 98 70 27 93	1.4 1.5 1.4 .3 1.4	60 40 50 20 60	7.0 10.3 11.4 9.7 8-4	ĸ
67 65 71 7.7 35	3.5 3.8 3.6 1.6 4.0	140 170 200 130 167	110 110 100 22 89	100 89 94 18 39	.1 .2 .1 .2	.14 <.09 <.10 .21 .12	<.01 .02 .02 .03	250 260 260 160 210	110 94 59 27 48	2.0 1.9 2.1 .3 1.1	50 40 50 20 50	7.2 9.8 11.9 9.9 7.3	Т, К
	_			_			_	_					
62 65 7.6 33	3.3 3.5 1.5 3.7	160 170 130 168	95 94 12 73	88 84 10 42	.1 .2 <.1 .2	<.10 <.10 .22 .10	<.02 <.01 .03 .01	240 250 150 220	83 81 24 52	1.9 2.1 .3 1.0	40 40 <.10 40	7.9 11.6 10-5 8.4	Т, К
	_			_		_	_		-				
		-		_									
								=					
150 150	6.3 5.6	170 180	170 160	200 180	.3 .3	.26 .14	.02	270 280	100 100	4.3 4.3	90 90	8.2 7.7	
30 27 27 4.5	2.8 2.7 2.7 1.4	150 160 170 130	85 83 83 11	35 33 31 6.4	.1 .2 .1	.16 <.09 <.10 .22	.02 <.01 <.10 .02	230 250 250 150	84 85 71 23	.9 .8 .8 .2	30 20 20 30	7.6 9.4 11.1 10.6	к
21 23	3.4 3.6	190 210	160 230	5.8 6.1	.1	.11 <.10	.01 .02	330 390	140 180	.6 .6	50 60	7.4 7.4	
-				—									
3.4	2.1	230	30	2.7	.1	.24	.03	270	37	.1	10	6.2	
2.3 8.1 8.6 4.8 5.4	1.2 1.5 1.4 1.2 1.5	210 180 210 190 177	<5.0 39 44 39 27	2.2 2.1 3.5 2.3 2.0	.1 .2 .2 .2	.11 .15 <.10 <.10	.01 .02 .02 <.01	250 230 240 230 200	43 54 34 37 26	.1 .3 .2 .2	10 20 20 20 20	7.5 8.0 8.2 8.1 7.0	F, K
3.0 2.2	1.4 .9	200 190	21 20	3.0 1.7	.2 .2	.13 <.10	.02 .01	230 220	25 31	.1 .1	0 10	8.2 8.6	
1.1 1.5 1.1	.5 .6 .4	92 130 104	<1.0 <5.0 10	2.0 1.8 1.0	.1 .1 <.1	.20 .13 .19	.02 .03 <.01	97 130 110	5 1 4	.1 0	0 0 <10	8.7 8.8 7.5	к
.9 1.2 1.0	.4 .3 .3	46 58 68	<5.0 <5.0 <5.0	.7 1.1 1.0	0 <.1 <.1	.12 .03 .10	.02 .03 <.01	54 57 78	8 0 10	.1 .1 .1	0 <10 <10	10 7 8.4 9.4	ĸ
26 28 23 28 31	4.4 5.1 4.2 3.7 5.4	150 170 180 160 125	140 140 110 130 260	19 18 18 20 17	.1 .3 .3 .3	1.0 .83 .69 .65 .88	<.02 <.01 <.01 .01 .01	270 290 260 270 360	120 120 81 110 230	.8 .8 .7 .8 .8	50 40 40 60 70	7.1 10-6 12.5 8.2 7.4	Т, К
							_	-					
41 27 14 22 32	8.3 5.5 3.6 3.8 6.2	200 300 240 260 306	390 230 120 200 280	19 9.9 6.7 8.8 12	.2 .4 .5 .4 .4	2.0 .74 .29 .31 1.2	.02 <.01 <.01 <.01 <.01 <.01	590 510 380 430 540	390 210 140 170 230	.8 .6 .4 .5 .7	140 70 30 50 100	6.5 8.4 10.8 9.3 8.1	т, р, к

Table 8 .--- Chemical analyses and suspended-sediment loads of

Site No	Site Name	Date of collection	Discharge (ft ³ /s)	Suspended sediment (tons/d)	Temperature (^O C)	рH	Specific conductance (umho)	Dissolved solids. sum of constituents (mg/L)	Dissolved silica (SiO ₂) (mg/L)	Dissolved calcium (Ca) (mg/L)	Dissolved magnesium (Mg) (mg/L)
61	Muddy Creek at mouth, Utah	8-20-81	Dry								
		5- 3-82	2.8		21.0	8.2	1,500	1,040	8.9	120	96
6.3	Dry Wash at mouth, Utah	8-20-81	Dry								
		5- 3-82	Dry								
64	East Fork Virgin River near	8-20-81	12	2.2	13.5	8.0	500	231	9.4	58	36
	Glendale, Utah	10-12-81	20		9.5	8.5	520	331	9.7	62	38
		2-12-82	17		2.5	8.1	500	300	8.6	52	36
		5-3-82	28	30	13.5	8.2	580	312	8.6	61	38
		8-31-82	13		14.0	8.2	500	266	9.4	54	25
65	Lydia's Canyon at mouth, Utah	8-19-81	Dry	~-							
	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	5- 3-82	2.9		16.0	8.0	540	346	9.3	64	43
68	Right Fork Lydia's Canyon at	8-19-81	1.0		16.0	8.3	460	250	8.8	47	37
	Roads end, Utah	5- 3-82	2.4		9.5	8.4	500	262	7.5	53	34
69	East Fork Virgin River above Lydia's Canyon, Utah	5 3-82	25	-	13.0	7.9	560	2 81	8.3	58	36
70	Stout Canyon at mouth, Utah	8-19-81	5.4		12.0	8.1	450	246	8.9	55	33
	······	5- 3-82	13		10.5	8.3	560	275	7.8	59	35
75	East Fork Virgin River above	8-19-81	5.0		12.5	8.4	440	244	8.9	50	33
	Stout Canyon, Utah	5- 3-82	7.3	-	10.0	8.2	550	286	8.6	58	36
77	East Fork Virgin River at	8-19-81	1.8	-	10.0	7.5	465	247	9.0	56	33
	Highway 136 bridge, Utah	10-12-81	2.1	_	8.5	8.2	470	263	8.9	57	34
	· · · · · · · · · · · · · · · · · · ·	2-11-82	1.4		9.5	7.8	450	273	8.2	55	32
		5-3-82	2.0	.55	10.0	8.0	475	284	8,9	58	33
		8-31-82	1.5		12.0	7.7	460	257	8.1	56	33

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water samples collected in the Virgin River basin, 1941-82.--Continued

Dissolved sodium (Na) (mg/L)	Dissolved potassium (K) (mg/L)	Alkalinity (total as CaCo ₃) (mg/L)	Dissolved sulfate (SO4) (mg/L)	Dissolved chloride (Cl) (mg/L)	Dissolved fluoride (F) (mg/L)	Dissolved nitrate (NO ₃)+ nitrate (NO ₂) as M (mg/L)	Phosphorus, ortho- phosphate as P (mg/L)	Total hardness (as CaCO ₃) (mg/L)	Non- carbonate hardness (as CaCo3) (mg/L)	Sodium- absorption ratio	Dissolved boron (B) (ug/L)	Dissolved oxygen (mg/L)	Other data availabie
 65	5.5	230	590	11	.3	<.10	<.01	700	470	1,2	70	7.7	
05	5.5												
								070	100	,	30	8-1	т, к
5.6	2.3	170	15 30	3.9	.4	.12	<.02 <.01	270 310	100 31	.1	30 20	8-1 8.6	1 , K
4.9 4.8	2.6	280 270	30 2.8	14 5.0	.5 .5	<.10	<.01	280	8.0	.1	20	11.5	
4.3	1.9	270	30	5.2	.5	<.10	<.01	310	39	.1	20	8.9	
4.8	2.3	223	22	4.3	.5	<.10	<.01	2 80	56	.1	30	8.1	
4.2	26	290	44	4.1	.6	<.10	.02	340	47	•1	30	7.1	
1.8	1.2	250	20	2.5	.1	.11	.01	260	13	.1	10	8.2	
1.9	.6	260	6	2.0	.2	<.10	.02	270	12	.1	<10	8.7	
4.3	1.5	260	11	5.5	.4	<.10	.01	2 90	33	.1	20	9.0	
2.1	1.4	220	9.0	1.8	.3	.13	.02	270	47	.1	10	8.5	
2.1	.9	260	11.0		.3	<.10	<.01	2 90	31	.1	10	8.9	
3.6	1.6	250	<5.0	5.2	.4	.14	.02	260	11	.1	10	8.9	
4.9	1.4	270	6.0		.4	.10	.01	2 90	23	.1	20	9.3	
2.6	1.3	230	<1.0	5.7	.4	.23	.01	280	46	.1	20	8.9	к
3.1	1.2	250	<5.0		.5	.21	.01	2 80	26	.1	20	8.5	
2.5	1.4	270	5.0		.6	.17	<.01	270	0	.1	20	9.1	
3.9 2.8	$1.2 \\ 1.3$	270	6.0		.5	.11	<.01 <.01	280 280	11	.1	20	8.6	
2.0	1.3	232	10	5.6	• 2	.12	<.01	280	44	.1	30	8.6	

Table 9.--Descriptive statistics of water-quality data at sites 1, 17, and 32 [Coefficient of variance: A unitless measure of variability calculated by the formula,

Standard deviation _____ X 100.] Mean

Mobreviations used: ft³, cubic feet per second; unho, micromhos; mm of hg, millimeters of mercury; ^oC, degrees Celsius; JTU, Jackson Tubidity Unit; FTU, Formazin Turbidity Unit; mg/L, milligrams per liter; UM-MF, micrometers-membrane filter; col/100 mL, colonies per 100 milliliters; KF agar, a base for culture media; PGi/L as K₄O, Pico curies per liter; fet-fld, fixed end point-field determination; acre-ft, acre feet; ug/L, micrograms per liter; cells/mL, cells per milliliter

Parameter	Number of analyses	Mean	Standard deviation	Minimum value	Maximum value	Coefficient of variance
	Sit	e l				
Discharge (ft ³ /s)	716	393.89	1342.27	40.00	27200.00	340.77
Streamflow, instantaneous (ft ³ /s)	300	540.14	982.38	50.00	10600.00	181.88
Specific conductance (umho)	686	2705.28	712.26	268.00	4650.00	26.33
pH (units)	612	8.03	2.15	7.00	27.00	26.75
Barometric pressure (mm of Hg)	18	713.61	18.43	651.00	741.00	2.58
Temperature, air (°C)	14	19.89	8.75	9.50	41.00	43.99
Temperature (°C)	360	18.34	6.31	5.00	30.60	34.42
Turbidity (JTU)	21	565.10	755.83	20.00	3100.00	133.75
Turbidity (FTU)	4	35.82	33.42	6.30	80.00	93.29
Oxygen, dissolved (mg/L)	65	8.91	1.30	6.60	13.00	14.58
Oxygen, dissolved (percent saturation)	15	83.96	30.95	6.60	103.00	36.86
Coliform, fecal, 0.45 UM-MF (cols/100 mL)	4	958.50	988.89	14.00	2000.00	103.17
Coliform, fecal, 0.7 UM-MF (cols/100 mL)	21	618.62	747.72	30.00	3000.00	120.87
Streptococci, fecal, KF agar (cols/100 mL)	20	3968.40	4813.23	108.00	15000.00	121.29
Streptococci, fecal, (cols/100 mL)	5	11095.98	19639.70	110.00	45999.91	177.00
Hardness (mg/L as CaOO ₃)	607	1051.58	347.08	27.00	2250.00	33.00
Hardness, noncarbonate (mg/L CaOO ₃)	611	971.15	603.65	120.00	3470.00	62.16
Calcium, dissolved (mg/L as Ca)	412	304.24	81.41	79.00	685.00	26.76
Magnesium, dissolved (mg/L as Mg)	411	97.18	41.39	18.00	313.00	42.59
Sodium, dissolved (mg/L as Na)	512	236.21	62.48	34.00	443.00	26.45
Percent sodium	454	30.77	5.47	7.00	46.00	21.03
Sodium-adsorption ratio	566	3.11	.67	.60	5.50	21.55
Sodium + potassium dissolved (mg/L as Na)	108	307.42	270.99	49.00	1170.00	88.15
Potassium 40 dissolved (pCi/L as K ₄ O)	8	12.95	4.09	6.90	21.00	31.58
Potassium, dissolved (mg/L as K)	344	23.45	6.60	3.30	47.00	28.15
Carbonate fet-fld $(mg/L \text{ as } OO_3)^{\circ}$	384	.11	1.38	.00	20.00	1266.91
Alkalinity field $(mg/L \text{ as } CaO_3)^{\circ}$	261	243.71	45.77	52.00	425.00	18.78
Carbon dioxide dissolved $(mg/L \text{ as } OO_2)^{\circ}$	138	9.96	8.76	.00	48.00	87.96
Sulfate, dissolved $(mg/L \text{ as } SO_4)^{\circ}$	609	862.65	323.67	.90	1960.00	37.52
Chloride, dissolved $(mg/L \text{ as } Cl)^{\circ}$	609	353.71	287.74	43.00	2460.00	81.35
Fluoride, dissolved (mg/L as F)	152	.82	.30	.20	1.90	36.34
Bromide dissolved (mg/L as BR)	113	.45	.24	.18	1.00	54.15
Silica, dissolved (mg/L as SiO ₂)	399	22.26	15.35	8.70	128.00	68.97
Solids, residue at 180°C, dissolved	507	1990.81	581.89	477.00	4250.00	29.23
Solids, sum of constituents, dissolved (mg/L)	211	1932.80	595.89	397.00	2890.00	30.83
Solids, dissolved (tons per acre-ft)	605	27.37	174.51	0.59	1410.00	637.51
Solids, dissolved (tons per day)	599	1055.53	1043.07	260.00	12000.00	98.82
Nitrogen, nitrate, total (mg/L as N)	11	.58	.30	.11	.96	50.79
Nitrogen, nitrate, dissolved (mg/L as N)	57	.47	.23	.05	.93	47.95
Nitrogen, nitrate, total (mg/L as NO3)	207	2.17	1.03	.20	5.60	47.54
Nitrogen, nitrate, total (mg/L as N) Nitrogen, nitrate, dissolved (mg/L as N) Nitrogen, nitrate, dissolved (mg/L as NO ₂) Nitrogen, NO ₂ + NO ₃ , total (mg/L as N) Nitrogen, NO ₂ + NO ₃ , dissolved (mg/L as N)	14 16 16 18 75	.02 .01 .03 .55 .45	.01 .00 .03 .28 .29	.01 .01 .12 .00	.04 .03 .10 .98 1.40	58.85 46.70 78.67 51.77 64.21
Nitrogen, ammonia, total (mg/L as N)	14	.08	.07	.01	.22	81.20
Nitrogen, ammonia, dissolved (mg/L as N)	14	.08	.06	.00	.21	72.02
Nitrogen, ammonia, dissolved (mg/L as NH4)	14	.11	.08	.00	.27	71.80
Nitrogen, ammonia + organic, dissolved (mg/L a	s N) 14	.65	.35	.28	1.70	53.47
Nitrogen, organic, total (mg/L as N)	12	1.56	1.59	.41	5.90	102.10
Nitrogen, organic dissolved (mg/L as N) Nitrogen, NH4 + organic suspended, total (mg/L as N)	14 14	.57 1.03	.34 1 .2 5	.24 .00	1.60 4.40	60.89 121.62

Mahla O Description	atotiatian of	wator-avality	data at ci	itos 1	17, and 32Continued
Table 9Descriptive	statistics of	water-quarter	uala al si	TCCD T	17, und 52 concentace

Parameter	Number of analyses	Mean	Standard deviation	Minimum value	Maximum value	Coefficient of variance	
Site 1Continued							
Nitrogen, ammonia + organic total (mg/L as N)	17	1.17	0.83	0.18	2.80	70.48	
Nitrogen, total (mg/L as NO3)	18	8.84	6.69	1.40	31.00	75.70	
Phosphorus, total (mg/L as P)	18	.66	1.24	.01	5.20	188.33	
Phosphorus, total (mg/L as PO4)	14	2.00	4.19	.00	16.00	209.17	
Phosphorus, dissolved (mg/L as P)	30	.03	.02	.01	.06	61.47	
Phosphorus, ortho, dissolved (mg/L as P)	50	.03	.02	.00	.13	&2.37	
Phosphorus, ortho, dissolved (mg/L as PO ₄)	74	.07	.07	.00	.40	103.54	
Arsenic, total (ug/L as As)	5	9.80	2.28	7.00	13.00	23.27	
Arsenic, suspended total (ug/L as As)	3	4.33	1.53	3.00	6.00	35.25	
Arsenic, dissolved (ug/L as As)	26	6.12	1.97	1.00	9.00	32.15	
Barium, dissolved (ug/L as Ba)	26	157.31	98.94	90.00	400.00	62.89	
Boron, total recoverable (ug/L as B)	2	4040.00	5600.28	80.00	8000.00	138.62	
Boron, dissolved (ug/L as B)	266	769.52	255.20	.31	1400.00	33.16	
Cadmium, total recoverable (ug/L as Cd)	3	7.33	11.02	.00	20.00	150.21	
Cadmium, suspended recoverable (ug/L as Cd)	3	4.00	5.29	.00	10.00	132.29	
Cadmium, dissolved (ug/L as Cd) Chromium, total recoverable (ug/L as Cr) Chromium, suspended recoverable (ug/L as Cr) Chromium, dissolved (ug/L as Cr) Cobalt, total recoverable (ug/L as Co)	6 5 5 8 5	3.33 14.00 8.00 8.50 23.80	8.16 13.42 10.95 9.90 42.86	.00 .00 .00 .00	20.00 30.00 20.00 20.00 100.00	244.95 95.83 136.93 116.46 180.10	
Cobalt, suspended recoverable (ug/L as Co) Cobalt, dissolved (ug/L as Co) Copper, total recoverable (ug/L as Cu) Copper, suspended recoverable (ug/L as Cu) Copper, dissolved (ug/L as Cu)	5 5 5 25	13.60 .40 24.60 20.00 2.12	20.84 .89 16.90 19.66 1.90	.00 .00 2.00 .00 .00	50.00 2.00 47.00 47.00 8.00	153.23 223.61 68.72 98.30 89.62	
Iron, total recoverable (ug/L as Fe)	16	2688.75	7246.29	.00	26000.00	269.50	
Iron, suspended recoverable (ug/L as Fe)	3	5080.00	8590.98	110.00	15000.00	169.11	
Iron, dissolved (ug/L as Fe)	29	28.96	20.76	10.00	90.00	71.68	
Iron, (ug/L as Fe)	12	72.50	192.08	.00	680.00	264.93	
Lead, total recoverable (ug/L as Pb)	3	82.33	103.23	7.00	200.00	125.38	
Lead, suspended recoverable (ug/L as Pb)	3	48.67	47.25	7.00	100.00	97.08	
Lead, dissolved (ug/L as Pb)	17	61.70	171.72	.00	600.00	278.29	
Manganese, total recoverable (ug/L as Mn)	5	474.00	571.86	30.00	1100.00	120.65	
Manganese, suspended recoverable (ug/L as Mn)	5	460.00	584.51	10.00	1100.00	127.07	
Manganese, dissolved (ug/L as Mn)	28	29.28	33.01	.00	180.00	113.03	
Mercury, total recoverable (ug/L as Hg)	5	.18	.18	.10	.50	99.38	
Mercury, suspended recoverable (ug/L as Hg)	5	.02	.04	.00	.10	223.61	
Mercury, dissolved (ug/L as Hg)	5	.18	.14	.10	.50	79.36	
Selenium, total (ug/L as Se)	5	1.80	.45	1.00	2.00	24.84	
Selenium, suspended total (ug/L as Se)	5	.20	.45	.00	1.00	223.61	
Selenium, dissolved (ug/L as Se)	26	1.46	.51	1.00	2.00	34.78	
Zinc, total recoverable (ug/L as Zn)	5	70.00	56.57	20.00	140.00	80.81	
Zinc, suspended recoverable (ug/L as Zn)	5	54.00	61.07	.00	130.00	113.10	
Zinc, dissolved (ug/L as Zn)	26	22.42	9.24	.00	40.00	41.23	
Phytoplankton, total (cells/mL)	6	815.00	982.79	140.00	2400.00	120.59	

Table 9.--Descriptive statistics of water-quality data at sites 1, 17, and 32--Continued

Parameter	Number of analyses	Mean	Standard deviation	Minimum value	Maximum value	Coefficient of variance	
SITE 17							
Discharge (ft ³ /s) Streamflow, (ft ³) Specific conductance (umho) pH (units) Barometric pressure (mm of Hg)	138 74 91 5 5	695.06 189.21 2049.34 7.96 683.60	1370.16 158.26 834.63 .09 7.76	48.00 40.00 270.00 7.90 674.00	8949.98 905.00 3999.99 8.10 695.00	197.13 83.64 40.73 1.12 1.14	
Temperature, air (^{O}C) Temperature (^{O}C) Oxygen, dissolved (mg/L) Hardness (mg/L as CaO_3) Hardness, noncarbonate (mg/L CaO_3)	4 205 4 5 1	18.12 16.53 8.48 778.00 352.00	13.24 8.07 1.59 537.70	.00 3.00 6.60 300.00 352.00	28.00 85.00 10.30 1700.00 352.00	73.07 48.82 18.74 69.11	
Calcium, dissolved (mg/L as Ca) Magnesium, dissolved (mg/L as Mg) Sodium, dissolved (mg/L as Na) Percent sodium Sodium-adsorption ratio	5 5 4 5	242.80 43.00 218.00 36.75 3.74	214.39 14.00 124.53 16.80 2.08	86.00 21.00 65.00 16.00 1.60	620.00 59.00 400.00 54.00 6.50	88.30 32.56 57.12 45.72 55.72	
Potassium, dissolved (mg/L as K) Carbonate fet-fld (mg/L as O_3) Sulfate, dissolved (mg/L as SO_4) Chloride, dissolved (mg/L as Cl) Fluoride, dissolved (mg/L as F)	5 1 5 5 5	16.72 .00 605.20 311.40 .42	7.71 523.50 185.08 .19	5.60 .00 140.00 87.00 .20	27.00 .00 1500.00 590.00 .70	46.09 86.50 59.43 45.80	
Silica, dissolved (mg/L as SiO ₂) Solids, residue at 180 ^O C dissolved (mg/L) Solids, sum of constituents, dissolved (mg/L) Solids, dissolved (tons per ac-ft) Solids, dissolved (tons per day)	5 1 4 5 5	15.38 1400.00 1626.00 2.15 709.00	5.75 920.88 1.09 260.00	9.90 1400.00 524.00 .71 439.00	23.00 1400.00 2670.00 3.63 1120.00	37.39 56.64 50.94 37.94	
Nitrogen, NO2+NO3, dissolved (mg/L as N) Phosphorus, ortho, dissolved (mg/L as P) Phosphate, ortho, dissolved (mg/L as PO4) Boron, dissolved (ug/L as B)	4 2 5	.56 .02 .09 476.00	.23 .01 .04 297.62	.30 .01 .06 140.00	.85 .04 .12 950.00	41.38 70.71 47.14 62.53	
	SIT	E 32					
Discharge (ft ³ /s) Streamflow, instantaneous (ft ³ /s) Specific conductance (umho) pH (units) Barometric pressure (mm of Hg)	171 41 42 7 4	806.13 175.95 787.00 8.08 665.75	1863.40 128.46 138.10 .21 7.93	48.00 76.00 430.00 7.80 658.00	9220.02 620.00 1000.00 8.30 674.00	231.15 73.01 17.55 2.62 1.19	
Temperature, air (^O C) Temperature (^O C) Oxygen, dissolved (mg/L) Hardness (mg/L as CaCO ₃) Hardness, noncarbonate (mg/L CaCO ₃)	4 173 4 7 3	23.62 17.00 8.28 365.43 297.33	8.09 7.24 1.56 88.38 84.36	18.50 1.50 6.30 190.00 201.00	35.50 29.50 10.10 456.00 358.00	34.23 42.58 18.84 24.19 28.37	
Calcium, dissolved (mg/L as Ca) Magnesium, dissolved (mg/L as Mg) Sodium, dissolved (mg/L as Na) Percent sodium Sodium-adsorption ratio	7 7 7 4 7	100.29 28.00 43.00 19.75 1.00	30.38 9.31 16.86 6.55 .39	55.00 13.00 10.00 10.00 .30	143.00 42.00 61.00 24.00 1.30	30.30 33.25 39.21 33.17 38.73	
Potassium, dissolved (mg/L as K) Carbonate fet-fld (mg/L as O_3) Sulfate, dissolved (mg/L as SO_4) Chloride, dissolved (mg/L as Cl) Fluoride, dissolved (mg/L as F)	5 3 7 7 5	3.90 .00 248.43 37.98 .20	1.14 .00 123.73 27.11 .07	2.00 .00 55.00 8.60 .10	5.00 .00 402.00 66.00 .30	29.24 49.80 71.37 35.36	
Silica, dissolved (mg/L as SiO ₂) Solids, residue at 180°C, dissolved (mg/L) Solids, sum of constituents, dissolved (mg/L) Solids, dissolved (tons per acre-ft) Solids, dissolved (tons per day)	5 3 4 7 7	9.56 699.67 472.25 .77 3542.32	2.32 14.47 162.67 .23 6578.78	7.20 683.00 238.00 .32 108.00	12.00 709.00 598.00 .96 17650.23	24.32 2.07 34.45 29.44 185.72	
Nitrogen, NO_2+NO_3 , dissolved (mg/L as N) Phosphorus, ortho, dissolved (mg/L as P) Phosphate, ortho, dissolved (mg/L as PO ₄) Boron, dissolved (ug/L as B)	4 4 2 5	.44 .02 .09 60.00	.15 .01 .00 21.21	.26 .01 .09 30.00	.62 .03 .09 80.00	33.86 57.74 .00 35.36	

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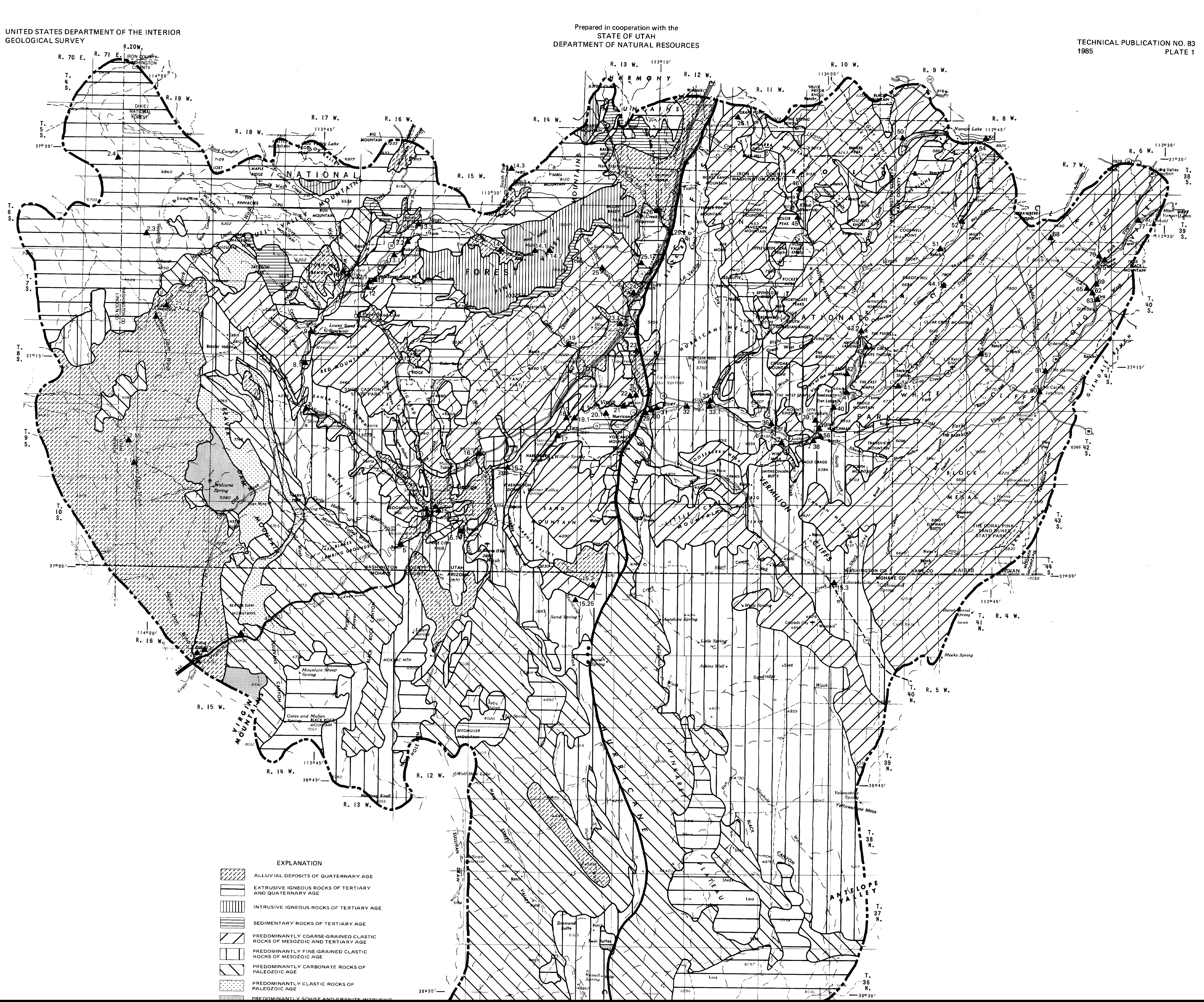
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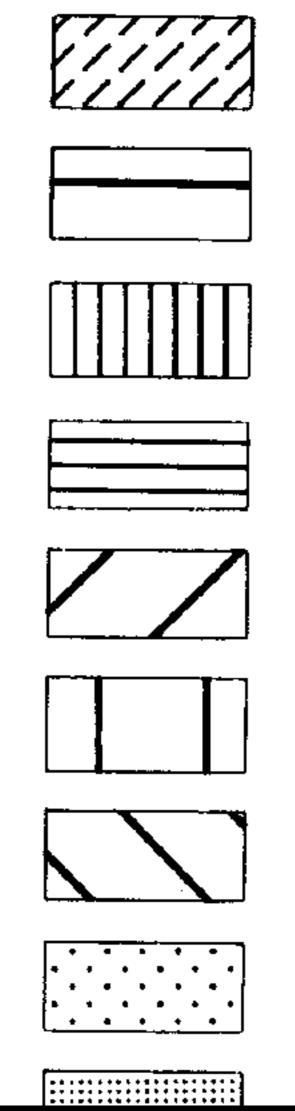
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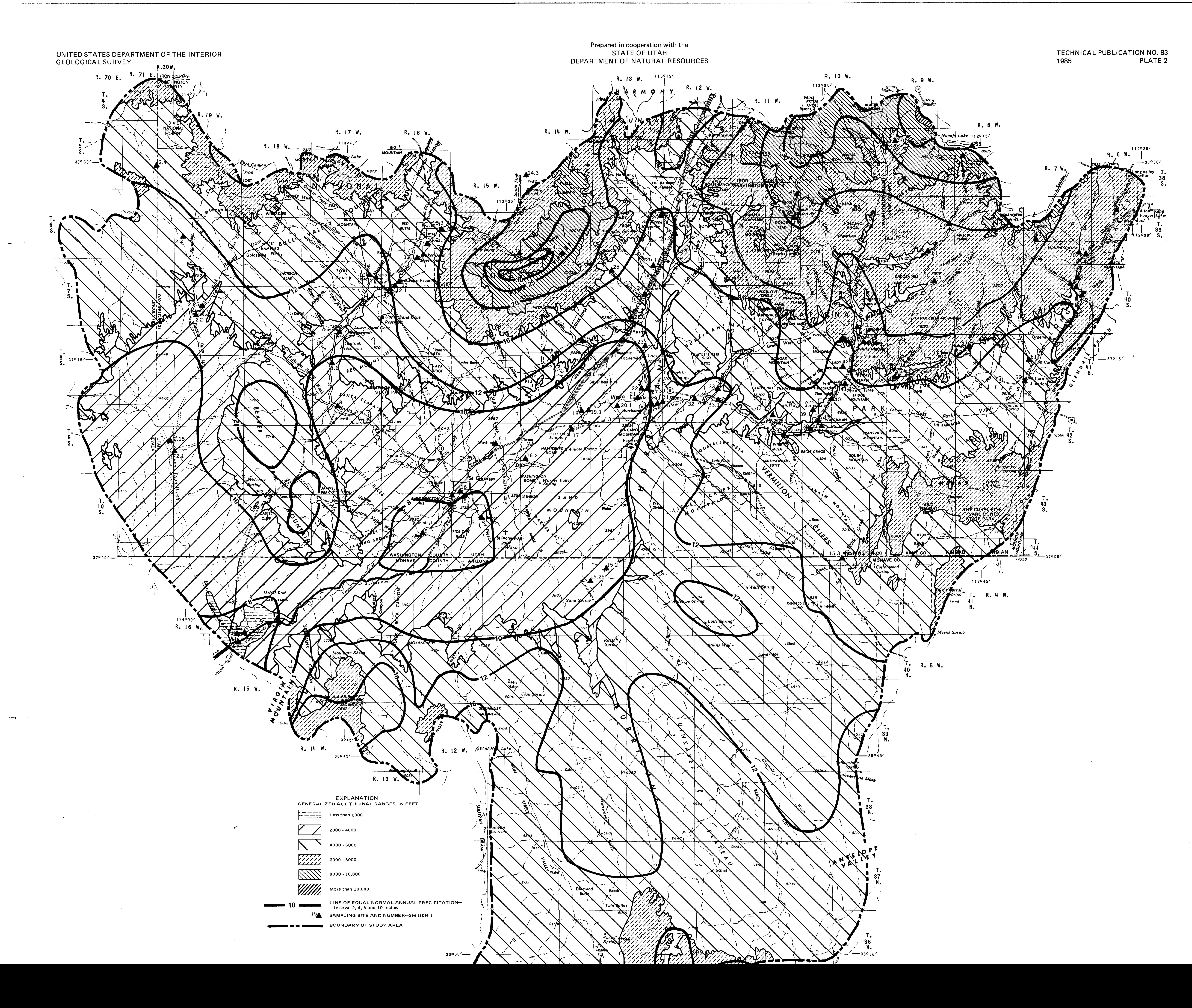
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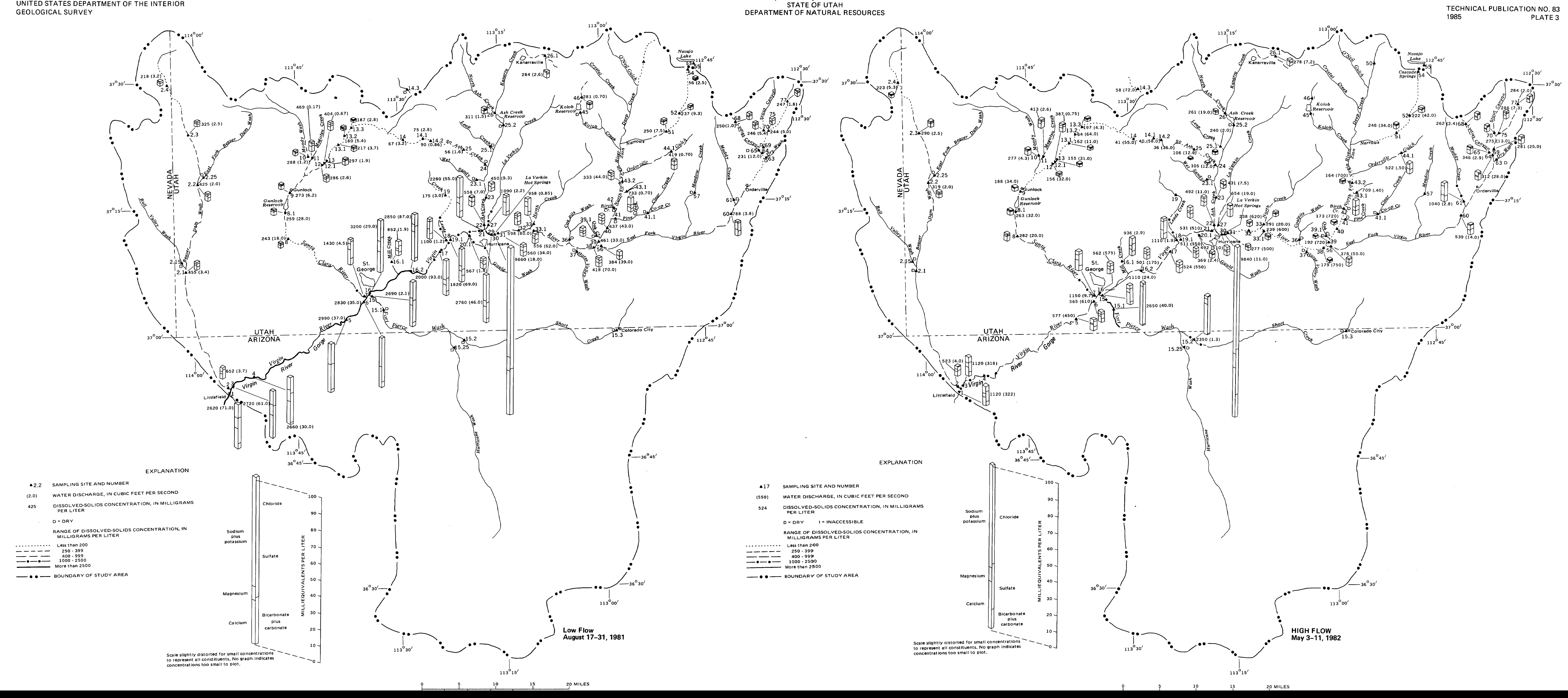
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